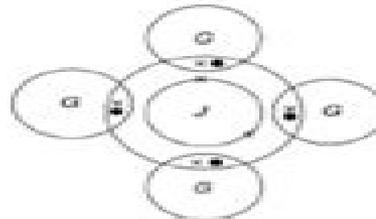


Chapter 5: Chemical Bonds

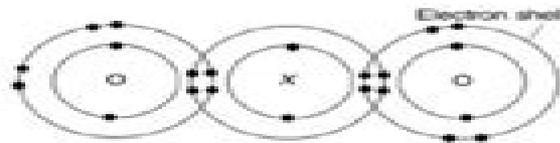
Paper 1

- 1) The diagram below shows the electron arrangement for the compound that is formed between atom *J* and atom *G*.



Which of the following statements is true about the compound?

- A The compound conducts electricity
 - B The compound is an ionic compound
 - C The melting point of the compound is low
 - D The atoms within the compound are attracted together by electrostatic force
- 2) What is the electron arrangement of atom *X*?



- A 2.4
 - B 2.6
 - C 2.8.6
 - D 2.8.4
- 3) The table below shows the information regarding two types of particles.

Particle	Number of protons	Electron configuration
<i>P</i>	8	2.8
<i>Q</i>	9	2.8

Both particles *P* and *Q* are

- A halide ions
- B noble gases
- C cations
- D anions

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 - The Rise of Digital Reading Chemistry Spm Question With Answer Full Download
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Spm Question With Answer Full Download
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Chemistry Spm Question With Answer Full Download
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Spm Question With Answer Full Download
 - Personalized Recommendations

- Chemistry Spm Question With Answer Full Download User Reviews and Ratings
- Chemistry Spm Question With Answer Full Download and Bestseller Lists
- 5. Accessing Chemistry Spm Question With Answer Full Download Free and Paid eBooks
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Spm Question With Answer Full Download
- 9. Balancing eBooks and Physical Books Chemistry Spm Question With Answer Full Download
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Spm Question With Answer Full Download
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Spm Question With Answer Full Download
 - Setting Reading Goals Chemistry Spm Question With Answer Full Download
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Spm Question With Answer Full Download
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- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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