

Chapter 17: Chemical Equilibrium

Worksheet: K_c and K_p

1. For the reaction $2\text{NO}_2(g) \rightleftharpoons \text{N}_2\text{O}_4(g)$ at 25°C, $K_c = 5.0$.

- 2. Calculate K_p for the reaction $2\text{NO}_2(g) \rightleftharpoons \text{N}_2\text{O}_4(g)$ at 25°C.
- 3. For the reaction $\text{N}_2(g) + 3\text{H}_2(g) \rightleftharpoons 2\text{NH}_3(g)$ at 25°C, $K_c = 0.50$. Calculate K_p .
- 4. For the reaction $\text{N}_2(g) + \text{O}_2(g) \rightleftharpoons 2\text{NO}(g)$ at 25°C, $K_c = 4.0$. Calculate K_p .
- 5. For the reaction $2\text{SO}_2(g) + \text{O}_2(g) \rightleftharpoons 2\text{SO}_3(g)$ at 25°C, $K_c = 1.0$. Calculate K_p .
- 6. For the reaction $\text{CO}_2(g) + \text{H}_2(g) \rightleftharpoons \text{CO}(g) + \text{H}_2\text{O}(g)$ at 25°C, $K_c = 1.0$. Calculate K_p .
- 7. For the reaction $\text{H}_2(g) + \text{I}_2(g) \rightleftharpoons 2\text{HI}(g)$ at 25°C, $K_c = 50.0$. Calculate K_p .
- 8. For the reaction $\text{H}_2(g) + \text{Cl}_2(g) \rightleftharpoons 2\text{HCl}(g)$ at 25°C, $K_c = 4.0$. Calculate K_p .
- 9. For the reaction $\text{H}_2(g) + \text{Br}_2(g) \rightleftharpoons 2\text{HBr}(g)$ at 25°C, $K_c = 16.0$. Calculate K_p .
- 10. For the reaction $\text{H}_2(g) + \text{S}_2(g) \rightleftharpoons 2\text{HS}(g)$ at 25°C, $K_c = 1.0$. Calculate K_p .

Answers

- 1. $K_p = 0.010$
- 2. $K_p = 0.010$
- 3. $K_p = 0.010$
- 4. $K_p = 0.010$
- 5. $K_p = 0.010$
- 6. $K_p = 1.0$
- 7. $K_p = 50.0$
- 8. $K_p = 4.0$
- 9. $K_p = 16.0$
- 10. $K_p = 1.0$

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Yi-Tong Ma



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Elements Of Language First Course Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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