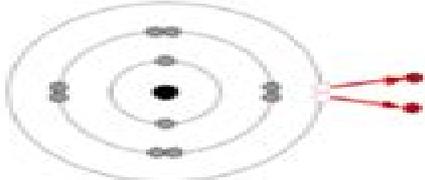
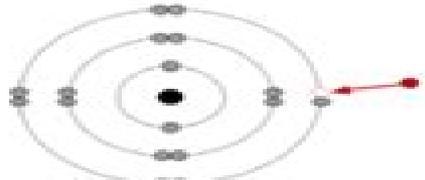


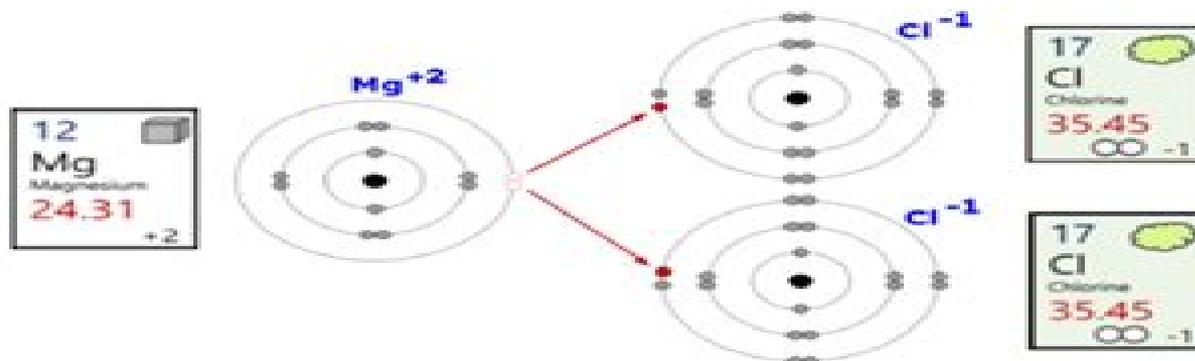
Ionic Bonding Puzzle Activity

Name: _____

When atoms of different elements bond together, they produce a compound. One category of compounds, known as **Ionic Compounds**, is formed when atoms transfer electrons to other atoms and become charged. The charged atoms are called **ions** and must combine in the correct ratio so that the overall charge is neutral (charge = 0).

		<p>Magnesium starts with 12 electrons and then <u>loses</u> 2 to make a complete outer energy shell</p>	<table border="1"> <tbody> <tr> <td>Protons = 12</td> </tr> <tr> <td>Electrons = 10</td> </tr> <tr> <td>Charge = +2</td> </tr> </tbody> </table>	Protons = 12	Electrons = 10	Charge = +2
Protons = 12						
Electrons = 10						
Charge = +2						
		<p>Chlorine starts with 17 electrons and then <u>gains</u> 1 to make a complete outer energy shell</p>	<table border="1"> <tbody> <tr> <td>Protons = 17</td> </tr> <tr> <td>Electrons = 18</td> </tr> <tr> <td>Charge = -1</td> </tr> </tbody> </table>	Protons = 17	Electrons = 18	Charge = -1
Protons = 17						
Electrons = 18						
Charge = -1						

Example: During a chemical reaction between Magnesium and Chlorine, the following occurs:



This creates the neutral compound called Magnesium chloride. 1 atom of magnesium combines with 2 atoms of chlorine with the chemical formula $MgCl_2$.

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