

NUCLEAR CHEMISTRY

BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction (α emission, β emission, γ emission, positron emission, artificial transmutation, fission, or fusion) described.

	Type of Nuclear Reaction
1.) ${}^{42}_{19}\text{K} \rightarrow {}^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	1.)
2.) ${}^{239}_{94}\text{Pu} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$	2.)
3.) ${}^{235}_{92}\text{U} \rightarrow \underline{\hspace{2cm}} + {}^{231}_{90}\text{Th}$	3.)
4.) ${}^1_1\text{H} + {}^2_1\text{H} \rightarrow \underline{\hspace{2cm}}$	4.)
5.) ${}^6_3\text{Li} + {}^1_0\text{n} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$	5.)
6.) ${}^{27}_{13}\text{Al} + {}^4_2\text{He} \rightarrow {}^{30}_{15}\text{P} + \underline{\hspace{2cm}}$	6.)
7.) ${}^9_4\text{Be} + {}^1_1\text{H} \rightarrow \underline{\hspace{2cm}} + {}^4_2\text{He}$	7.)
8.) ${}^{37}_{19}\text{K} \rightarrow {}^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	8.)
9.) $\underline{\hspace{2cm}} + {}^1_0\text{n} \rightarrow {}^{142}_{56}\text{Ba} + {}^{91}_{36}\text{Kr} + 3 {}^1_0\text{n}$	9.)
10.) ${}^{238}_{92}\text{U} + {}^4_2\text{He} \rightarrow \underline{\hspace{2cm}} + {}^1_0\text{n}$	10.)
11.) ${}^{14}_6\text{C} \rightarrow {}^{14}_7\text{N} + \underline{\hspace{2cm}}$	11.)
12.) ${}^{187}_{75}\text{Re} + \underline{\hspace{2cm}} \rightarrow {}^{188}_{75}\text{Re} + {}^1_1\text{H}$	12.)
13.) ${}^{22}_{11}\text{Na} + \underline{\hspace{2cm}} \rightarrow {}^{22}_{10}\text{Ne}$	13.) ---
14.) ${}^{218}_{84}\text{Po} \rightarrow \underline{\hspace{2cm}} + {}^4_2\text{He}$	14.)
15.) ${}^{253}_{99}\text{Es} + {}^4_2\text{He} \rightarrow {}^1_0\text{n} + \underline{\hspace{2cm}}$	15.)

Write the Balanced nuclear equations for the alpha decay of:

a) Plutonium-234

b) Strontium-90

Write the balanced nuclear equations for the alpha, beta and gamma decay of Radium-226

Chapter 25 Nuclear Chemistry Worksheet

DP Hallahan



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