

Chemistry Chapter 4

Balancing chemical Equations

Chemical Reactions

- Reactions involve chemical changes in matter resulting in new substances
- Reactions involve rearrangement and exchange of atoms to produce new molecules
 - Elements are not transmuted during a reaction

Chemical Equations



- CH_4 and O_2 are the reactants, and CO_2 and H_2O are the products
- the (g) after the formulas tells us the state of the chemical
- the number in front of each substance tells us the numbers of those molecules in the reaction
 - called the **coefficients**

Balancing chemical equations

- Step 1: Determine the state of the reactants and products in the chemical reaction
- Step 2: Write the unbalanced chemical equation
- Step 3: Balance the number of atoms of each element in the reactant and product side, starting with the most complicated molecule

Balancing a Hypothetical Chemical Equation



- In this hypothetical reaction, C and D are the reactants, and X and Y are the products
- The number in front of each substance tells us the numbers of those molecules of reactants and products in the reaction
- So, if you react exactly 3 moles of C with 2 moles of D, then you will produce 5 moles of X and 2 moles of Y.
- If you reacted 6 moles of C with 4 moles of D, you will produce 10 moles of X and 4 moles of Y.

Example: Balancing chemical equations

- Glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) reacts with Oxygen gas to produce gaseous carbon dioxide and water vapor



Reactants

6 C

12 H

8 O

Products

1 C

2 H

3 O

Chapter 3 And 4 Chemistry Test

**Richard A. McPherson, Matthew R.
Pincus**



Chapter 3 And 4 Chemistry Test:

McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best This book includes 4 full length sample tests updated for the latest test format 40 top tips to remember on test day Glossary of tested chemistry terms and formulas Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School Diagnostic test to pinpoint strengths and weaknesses Step by step review of all topics covered on the exam In depth coverage of the lab experiment questions that are a major test feature Charts tables and illustrations to simplify and reinforce learning Practice tests just like the real SAT Subject Test in Chemistry Test taking tips and strategies McGraw-Hill's SAT Subject Test: Chemistry, 2ed Thomas Evangelist, 2009-02-01 We want to help you score high on the SAT Chemistry test We've put all of our proven expertise into McGraw Hill's SAT Subject Test Chemistry to make sure you're fully prepared for this difficult exam With this book you'll get essential skill building techniques and strategies created by a leading high school chemistry teacher You'll also get 5 full length practice tests hundreds of sample questions and all the facts about the current exam With McGraw Hill's SAT Subject Test Chemistry we'll guide you step by step through your preparation program and give you the tools you need to succeed 5 full length sample tests with complete explanations for every question 40 top test items to remember on exam day A step by step review of all topics covered on the exam Teacher recommended tips and strategies to help you raise your score

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Standard Methods of the Division of Laboratories and Research of the New York State Department of Health Augustus Baldwin Wadsworth,1927

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A Guide to Educational Measurements Harlan Cameron Hines, 1923 **An Evaluation of the Metabolic Profile Test Predictive Value for Post Partum Disease and Sources of Error in Field Applications of Clinical Chemistry and Hematology Tests in Dairy Cattle** Elden George Lamprecht, 1978 **The High School**, 1923 *Point-of-care testing* Peter Lupp, Ralf Junker, 2018-07-18 The underlying technology and the range of test parameters available are evolving rapidly The primary advantage of POCT is the convenience of performing the test close to the patient and the speed at which test results can be obtained compared to sending a sample to a laboratory and waiting for results to be returned Thus a series of clinical applications are possible that can shorten the time for clinical decision making about additional testing or therapy as delays are no longer caused by preparation of clinical samples transport and central laboratory analysis Tests in a POC format can now be found for many

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