

Introduction to Acid, Base, and Salt

- The nature of calcium phosphate present in tooth enamel is
a. Basic b. Amphoteric c. Acidic d. Neutral
Answer ____
- Which of the following salts has no water of crystallization?
a. Blue vitriol b. Washing soda c. Baking soda d. Gypsum
Answer ____
- The pH of an HCl solution is 4. What is the molarity of the solution?
a. 4.0 M b. 0.4 M c. 0.0001 M d. 0.001 M
Answer ____
- Which of the following does not form an acidic salt?
a. Phosphoric acid b. Carbonic acid
c. Hydrochloric acid d. Sulfuric acid
Answer ____
- The gas that is released when an acid reacts with a metal carbonate is
a. Carbon dioxide b. Hydrogen c. Oxygen d. Nitrogen
Answer ____
- Two fruits that are rich in citric acid are
a. Lemon and tamarind b. Lemon and orange
c. Orange and tamarind d. Tomato and tamarind
Answer ____
- The acid present in stinging hair of nettle leaves that causes burning pain
a. Citric acid b. Acetic acid c. Tartaric acid d. Methanoic acid
Answer ____
- As the pH value of a solution decreases,
a. Number of OH^- ions increases b. Number of H^+ ions increases
c. Number of H^+ ions decreases d. Equal number of H^+ ions and OH^- ions
Answer ____

Chapter 8 Solutions Acids Bases Worksheet

Richard Harwood, Ian Lodge



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