

Name: _____

83

(a) (10 Points) Write down the total energy balance equation for an open system. Define all terms.

$$\frac{d(nU)}{dt} + \Delta \left[\left(H + \frac{u^2}{2} + zg \right) \dot{n} \right]_{in} = \dot{Q} + \dot{W}$$

- n = no. of moles
- U = internal energy
- H = Enthalpy
- u = velocity (speed)
- z = height (relative position)
- \dot{n} = flow rate
- \dot{Q} = rate of flow of heat
- \dot{W} = rate of work done

$\Delta = final - initial$
 $i = \text{flow streams}$

Used to define PE / KE

(b) (10 Points) Explain how it is possible to change a vapor into a liquid without condensation.

You could decrease the volume and increase the pressure isothermally. This would mean that the temperature would remain constant while volume and pressure changes, changing a vapor into a liquid without condensation.

need to go above critical pt.

Chemical Reaction Engineering Final Exam Solution

Thomas Griffiths



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