

STUDY GUIDE

SECTION 1.1 - 1.4

- **matter**: everything that takes up space + has a mass

Properties of matter



① Physical changes:

- Phase
- Size
- Texture

→ Reversible

② Chemical changes:

- Color change
- New substance is formed

→ Irreversible

Pure substance vs Mixture

MOLECULE (pure substances)

Compounds

2 or more \neq atoms
(eg: H_2O ; Na_2O)

Elements

1 atom
(eg: H_2 , O_2)

- **Pure substance**: constant composition + definite properties

Chemistry Midterm Study Guide

Sabine Zange



Chemistry Midterm Study Guide:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **Chemistry Midterm Study Guide**

In a world inundated with displays and the cacophony of immediate communication, the profound energy and emotional resonance of verbal art usually disappear into obscurity, eclipsed by the regular assault of noise and distractions. However, nestled within the musical pages of **Chemistry Midterm Study Guide**, a charming work of literary elegance that pulses with organic thoughts, lies an unforgettable trip waiting to be embarked upon. Written with a virtuoso wordsmith, this interesting opus guides visitors on a mental odyssey, delicately revealing the latent possible and profound impact embedded within the complex web of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book's main themes, dissect their captivating publishing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers' souls.

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