

3 - 4 Review and Reinforcement

Elements and Compounds

Put each of the following elements in the chart:

Elemental Form	Element Symbol	Elemental Mass
1. Carbon	C	12.01
2. Hydrogen	H	1.01
3. Neon	Ne	20.18
4. Magnesium	Mg	24.31
5. Iron	Fe	55.85
6. Zinc	Zn	65.38
7. Potassium	K	39.10
8. Lead	Pb	207.2
9. Sodium	Na	22.99
10. Nitrogen	N	14.01

Complete the following by inserting the missing words appropriate to each of the lines:

element symbol periodic table
 mixture compound atom mass
 compound element matter matter matter

1. Elements are often organized by physical or chemical properties.
2. Elements are often grouped into metals, metalloids, and nonmetals.
3. Elements are organized by their periodic table.
4. The periodic table is usually divided into groups.
5. Properties are similar when elements are in the same group.
6. Elements and compounds are made from particles because they have a unique set of chemical and physical properties.

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