



# higher education & training

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REPUBLIC OF SOUTH AFRICA

## NATIONAL CERTIFICATE ENGINEERING SCIENCE II

[1200 FOM]

20 July 2020 (20 papers)  
09:00 – 12:00

Essential instruments and equipment (e.g. scientific calculator) may be used.

This question paper consists of 02 pages and 2 formula sheets.

1. INSTRUCTIONS

# N1 Engineering Science Question Paper

**United States. National Bureau of  
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## **N1 Engineering Science Question Paper:**

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*Papers Presented at the Twelfth International Conference on Fluid Sealing* B. S. Nau,1989 Very Good No Highlights or Markup all pages are intact      Publications of the National Bureau of Standards United States. National Bureau of Standards,1986      **Science Abstracts. Physics and Electrical Engineering** ,1909      **Science Abstracts** ,1913

**Current Index to Journals in Education** ,2000      **Science and Art of Mining** ,1916      **Ambient Intelligence** Juan Carlos Augusto,Reiner Wichert,Rem Collier,David Keyson,Albert A. Salah,Ah-Hwee Tan,2013-11-23 This book constitutes the refereed proceedings of the 4th International Joint Conference an Ambient Intelligence AmI 2013 held in Dublin Ireland in December 2013 The 15 revised full papers 4 papers from the landscape track 3 papers from the doctoral colloquium and 6 demo and poster papers were carefully reviewed and selected from numerous submissions and are presented with 6 workshop descriptions The papers cover a variety of multi disciplinary topics in computer science human computer interaction electrical engineering industrial design behavioral sciences distributed devices ubiquitous and communication technologies pervasive computing intelligent user interfaces and artificial intelligence      Publications of the National Bureau of Standards ,1986 Catalog United States. National Bureau of Standards,1987      Engineering ,1879      *Publications* United States. National Bureau of Standards,1987      *New Haven Free Public Library Bulletin* New Haven Free Public Library,1913      **The Papers of Woodrow Wilson** Woodrow Wilson,1972 This massive collection includes all important letters speeches interviews press conferences and public papers on Woodrow Wilson The volumes make available as never before the materials essential to understanding Wilson s personality his intellectual religious and political development and his careers as educator writer orator and statesman The Papers not only reveal the private and public man but also the era in which he lived making the series additionally valuable to scholars in various fields of history between the 1870 s and the 1920 s Publisher      **Power and the Engineer** ,1898      **Engineering Science N1** ,2000      **Serials Holdings** Linda Hall Library,1989      English Mechanic and Mirror of Science and Art ,1883

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this video accompanies the chm 252 lab nitration of bromobenzene in this video students will get an introduction to the lab as well as the set up for conducting the reaction students should

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho/para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical  $\times 100$  what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4 nitrobromobenzene and 2

nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion  $\text{NO}_2^+$  which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0ml of nitric acid and 4.0ml of sulfuric acid and will then react with 4.5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment

nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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