

AP Chemistry Lab 07

Investigation 3 *What Makes Hard Water Hard?*

Central Challenge

Samples of tap water have been collected from faculty members who live in different townships, villages, cities, and counties. Some are from municipal water sources, others are from wells. The water will be analyzed for their quantities of water hardness through principles of metal ion precipitation and separation. The samples will then be ranked in order of increasing water hardness.

Context for This Investigation

When relocating, there are many factors that go into choosing where to buy a house. One of those factors is water quality – it is advantageous to find a location where a water softener will not be needed. Hard water costs more for detergent, plumbing repairs, and appliance maintenance and replacement. Having water that is already soft is preferred to installing a water softening system. You, the students, will identify which of the areas around Western New York have the lowest hard water quantities by ranking the hardness of the water in the areas.

Materials

Distilled water	Metal scoops	250 mL Erlenmeyer flasks (optional)
Calcium chloride, anhydrous (for practice)	250 mL beakers	Buchner funnels with stoppers
Sodium carbonate, anhydrous (for practice)	100 mL graduated cylinders	Glass tubing (optional)
Wash bottles	Watch glasses	Ring stand
Analytical balances	250 mL side-arm flasks (filter flasks)	Drying oven
Clamps	Aspirators	Filter papers
Double hole rubber stoppers (optional)		Water samples

Safety and Disposal

All solutions can be flushed down the drain. Normal laboratory precautions, including wearing goggles at all times, should be taken as these solutions may be harmful if swallowed and can irritate the eyes.

Prelab Guiding Questions/Simulations

Ap Chemistry Lab Answers

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Table of Contents Ap Chemistry Lab Answers

1. Understanding the eBook Ap Chemistry Lab Answers
 - The Rise of Digital Reading Ap Chemistry Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Chemistry Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Chemistry Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Chemistry Lab Answers
 - Personalized Recommendations
 - Ap Chemistry Lab Answers User Reviews and Ratings
 - Ap Chemistry Lab Answers and Bestseller Lists
5. Accessing Ap Chemistry Lab Answers Free and Paid eBooks
 - Ap Chemistry Lab Answers Public Domain eBooks
 - Ap Chemistry Lab Answers eBook Subscription Services
 - Ap Chemistry Lab Answers Budget-Friendly Options

6. Navigating Ap Chemistry Lab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ap Chemistry Lab Answers Compatibility with Devices
 - Ap Chemistry Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ap Chemistry Lab Answers
 - Highlighting and Note-Taking Ap Chemistry Lab Answers
 - Interactive Elements Ap Chemistry Lab Answers
8. Staying Engaged with Ap Chemistry Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Chemistry Lab Answers
9. Balancing eBooks and Physical Books Ap Chemistry Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Chemistry Lab Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ap Chemistry Lab Answers
 - Setting Reading Goals Ap Chemistry Lab Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ap Chemistry Lab Answers
 - Fact-Checking eBook Content of Ap Chemistry Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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