

Name _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) According to kinetic-molecular theory, in which of the following gases will the root-mean-square speed of the molecules be the highest at 200°C?
- A) SF₆
 B) H₂O
 C) HCl
 D) Cl₂
 E) None. The molecules of all gases have the same root-mean-square speed at any given temperature.

SHORT ANSWER. Write the word or phrase that best completes each statement or answers the question.

- 2) What two properties of real gases cause deviation from ideal behavior?

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 3) A sample of a gas (5.0 mol) at 1.0 atm is expanded at constant temperature from 10 L to 15 L. The final pressure is _____ atm.
- A) 1.5 B) 15 C) 0.67 D) 3.3 E) 7.5
- 4) A sample of He gas (2.35 mol) occupies 57.9 L at 300.0 K and 1.00 atm. The volume of this sample is _____ L at 423 K and 1.00 atm.
- A) 0.709 B) 57.9 C) 41.1 D) 81.6 E) 1.41
- 5) The density of N₂O at 1.53 atm and 45.2°C is _____ g/L.
- A) 0.388 B) 2.58 C) 9.99 D) 1.76 E) 18.2
- 6) The volume of a sample of gas (2.49 g) was 752 mL at 1.98 atm and 62°C. The gas is _____.
- A) NO₂ B) SO₃ C) SO₂ D) Ne E) NH₃
- 7) What volume (mL) of sulfur dioxide can be produced by the complete reaction of 3.82 g of calcium sulfite with excess HCl (aq), when the final SO₂ pressure is 827 torr at 44°C?
- A) 1.39×10^{-4} B) 0.106 C) 1.00×10^{-3} D) 761 E) 578
- 8) The Mond process produces pure nickel metal via the thermal decomposition of nickel tetracarbonyl:



What volume (L) of CO is formed from the complete decomposition of 444 g of Ni(CO)₄ at 752 torr and 22°C?

- A) 255 B) 20.2 C) 63.7 D) 11.0 E) 0.356

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