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# **AQA a level physics paper 1 100% SOLUTIONS REVISED 2023//2024**

What is meant by work function (2) - ANSWER Minimum energy required by an electron to escape from (1) a metal surface (1)

What is meant by ionisation energy? (2) - ANSWER Minimum energy required to remove an electron from an atom (1) from the ground state (1)

Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4)  
- ANSWER Energy of a photon is fixed/constant (1)  
Energy is required for electrons to overcome work function  $\phi$  (1)  
Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE(1)  
Maximum KE = Energy of photon - work function  $\phi$  (1)

Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons? - ANSWER Diffraction pattern maxima/minima (bright and dark areas) (1)  
Diffraction is a property of waves- proves wave-behaviour of electrons (1)

What is meant by the ground state? - ANSWER The lowest energy state of an atom

Explain the difference between excitation and ionisation (2) - ANSWER An electron receives a discrete amount of energy for BOTH (1)  
Excitation promotes an electron to a higher energy level  
Ionisation is when an electron receives enough energy to leave the atom

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**VM Jensen**



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## **Table of Contents Aqa Physics Past Paper Answers**

1. Understanding the eBook Aqa Physics Past Paper Answers
  - The Rise of Digital Reading Aqa Physics Past Paper Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Aqa Physics Past Paper Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Aqa Physics Past Paper Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Aqa Physics Past Paper Answers
  - Personalized Recommendations
  - Aqa Physics Past Paper Answers User Reviews and Ratings

- Aqa Physics Past Paper Answers and Bestseller Lists
- 5. Accessing Aqa Physics Past Paper Answers Free and Paid eBooks
  - Aqa Physics Past Paper Answers Public Domain eBooks
  - Aqa Physics Past Paper Answers eBook Subscription Services
  - Aqa Physics Past Paper Answers Budget-Friendly Options
- 6. Navigating Aqa Physics Past Paper Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Aqa Physics Past Paper Answers Compatibility with Devices
  - Aqa Physics Past Paper Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Aqa Physics Past Paper Answers
  - Highlighting and Note-Taking Aqa Physics Past Paper Answers
  - Interactive Elements Aqa Physics Past Paper Answers
- 8. Staying Engaged with Aqa Physics Past Paper Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Aqa Physics Past Paper Answers
- 9. Balancing eBooks and Physical Books Aqa Physics Past Paper Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Aqa Physics Past Paper Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Aqa Physics Past Paper Answers
  - Setting Reading Goals Aqa Physics Past Paper Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aqa Physics Past Paper Answers
  - Fact-Checking eBook Content of Aqa Physics Past Paper Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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