

## Balancing Equations Worksheet

- 1) \_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_  $\text{KOH}$   $\rightarrow$  \_\_\_  $\text{NaOH}$  + \_\_\_  $\text{K}_3\text{PO}_4$
- 2) \_\_\_  $\text{MgF}_2$  + \_\_\_  $\text{Li}_2\text{CO}_3$   $\rightarrow$  \_\_\_  $\text{MgCO}_3$  + \_\_\_  $\text{LiF}$
- 3) \_\_\_  $\text{P}_4$  + \_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_  $\text{P}_2\text{O}_3$
- 4) \_\_\_  $\text{RbNO}_3$  + \_\_\_  $\text{BeF}_2$   $\rightarrow$  \_\_\_  $\text{Be}(\text{NO}_3)_2$  + \_\_\_  $\text{RbF}$
- 5) \_\_\_  $\text{AgNO}_3$  + \_\_\_  $\text{Cu}$   $\rightarrow$  \_\_\_  $\text{Cu}(\text{NO}_3)_2$  + \_\_\_  $\text{Ag}$
- 6) \_\_\_  $\text{CF}_4$  + \_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_  $\text{CBr}_4$  + \_\_\_  $\text{F}_2$
- 7) \_\_\_  $\text{HCN}$  + \_\_\_  $\text{CuSO}_4$   $\rightarrow$  \_\_\_  $\text{H}_2\text{SO}_4$  + \_\_\_  $\text{Cu}(\text{CN})_2$
- 8) \_\_\_  $\text{GaF}_3$  + \_\_\_  $\text{Cs}$   $\rightarrow$  \_\_\_  $\text{CsF}$  + \_\_\_  $\text{Ga}$
- 9) \_\_\_  $\text{BaS}$  + \_\_\_  $\text{PtF}_2$   $\rightarrow$  \_\_\_  $\text{BaF}_2$  + \_\_\_  $\text{PtS}$
- 10) \_\_\_  $\text{N}_2$  + \_\_\_  $\text{H}_2$   $\rightarrow$  \_\_\_  $\text{NH}_3$
- 11) \_\_\_  $\text{NaF}$  + \_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_  $\text{NaBr}$  + \_\_\_  $\text{F}_2$
- 12) \_\_\_  $\text{Pb}(\text{OH})_2$  + \_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_  $\text{H}_2\text{O}$  + \_\_\_  $\text{PbCl}_2$
- 13) \_\_\_  $\text{AlBr}_3$  + \_\_\_  $\text{K}_2\text{SO}_4$   $\rightarrow$  \_\_\_  $\text{KBr}$  + \_\_\_  $\text{Al}_2(\text{SO}_4)_3$
- 14) \_\_\_  $\text{CH}_4$  + \_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_  $\text{CO}_2$  + \_\_\_  $\text{H}_2\text{O}$
- 15) \_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_  $\text{CaCl}_2$   $\rightarrow$  \_\_\_  $\text{NaCl}$  + \_\_\_  $\text{Ca}_3(\text{PO}_4)_2$
- 16) \_\_\_  $\text{K}$  + \_\_\_  $\text{Cl}_2$   $\rightarrow$  \_\_\_  $\text{KCl}$
- 17) \_\_\_  $\text{Al}$  + \_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_  $\text{H}_2$  + \_\_\_  $\text{AlCl}_3$
- 18) \_\_\_  $\text{N}_2$  + \_\_\_  $\text{F}_2$   $\rightarrow$  \_\_\_  $\text{NF}_3$
- 19) \_\_\_  $\text{SO}_2$  + \_\_\_  $\text{Li}_2\text{Se}$   $\rightarrow$  \_\_\_  $\text{SSe}_2$  + \_\_\_  $\text{Li}_2\text{O}$
- 20) \_\_\_  $\text{NH}_3$  + \_\_\_  $\text{H}_2\text{SO}_4$   $\rightarrow$  \_\_\_  $(\text{NH}_4)_2\text{SO}_4$

# Balancing Equation Practice Packet Answer Sheet

**J Elliott**



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## **Balancing Equation Practice Packet Answer Sheet** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Balancing Equation Practice Packet Answer Sheet**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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