

CHAPTER 4 OBJECTIVES

# Chemical Equations and Reactions

## LEARNING OBJECTIVES

After studying this chapter, you should be able to:

1. (a) balance a chemical equation representing all the following through:
  - (i) experimentally established facts,
  - (ii) stoichiometric by which elements combine in their products,
  - (iii) identities of reactants and products in a chemical reaction,
  - (iv) relative quantities of reactants and products in a chemical reaction.
2. According to the law of conservation of mass, the total mass of the reacting substances is:
  - (a) always more than the total mass of the products,
  - (b) always less than the total mass of the products,
  - (c) sometimes more and sometimes less than the total mass of the products,
  - (d) always equal to the total mass of the products.
3. Explain why the total mass of the following chemical reactions will remain the same. Write the balanced equation for each reaction and explain the chemical equation writing in the products. Illustrate by drawing the final product. (Illustration may refer to Figure 4.3) in the text.



# Chapter 8 Chemical Reactions Guided Reading Answers

**O García**



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