

Thermodynamic Equilibrium - 1 (Application of Equilibrium)

Thermodynamic Equilibrium

Thermodynamic Equilibrium: A system is in thermodynamic equilibrium if its macroscopic properties do not change with time.

1. In a closed system at constant volume and pressure, which of the following is not a state function?
- A. Entropy
 - B. Heat
 - C. Work
 - D. All of the above
2. For the thermodynamic system shown below, which process is not a state function?
- A. Entropy
 - B. Heat
 - C. Work
 - D. All of the above
3. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above
4. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above
5. Which of the following is not a state function? (Select all that apply.)
- A. Entropy
 - B. Heat
 - C. Work
 - D. None of the above
6. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above
7. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above
8. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above
9. Which of the following is not a state function? (Select all that apply.)
- A. Heat
 - B. Entropy
 - C. Work
 - D. None of the above

Chapter Test Thermal Energy Answers 16

Judd E. Hollander



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