

Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

- | | |
|----------------------------------|-----------------------------------|
| <u> c </u> synthesis | (a) $AX + BY \rightarrow AY + BX$ |
| <u> d </u> decomposition | (b) $A + BX \rightarrow AX + B$ |
| <u> b </u> single-displacement | (c) $A + B \rightarrow AX$ |
| <u> a </u> double-displacement | (d) $AX \rightarrow A + X$ |

2. c In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$, iron has been replaced by
 (a) nitrate. (c) **aluminum.**
 (b) water. (d) nitrogen.

3. a Of the following chemical equations, the only reaction that is both synthesis and combustion is
 (a) **$C(s) + O_2(g) \rightarrow CO_2(g)$.**
 (b) $2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.
 (c) $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(aq) + 6O_2(g)$.
 (d) $C_6H_{12}O_6(aq) + 6O_2(g) \rightarrow 6CO_2(aq) + 6H_2O(l)$.

4. b Of the following chemical equations, the only reaction that is both combustion and decomposition is
 (a) $S(s) + O_2(g) \rightarrow SO_2(g)$.
 (b) **$2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.**
 (c) $2H_2O_2(l) \rightarrow 2H_2O(l) + O_2(g)$.
 (d)

5. Identify the products when the following substances decompose:

- Its separate elements** _____ a binary compound
- Metal oxide + water** _____ most metal hydroxides
- Metal oxide + carbon dioxide** _____ a metal carbonate
- Water + sulfur dioxide** _____ the acid H_2SO_3

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

Chemical Equations Reactions Section 2 Answers

Michael Seilmaier



Chemical Equations Reactions Section 2 Answers:

Unveiling the Magic of Words: A Review of "**Chemical Equations Reactions Section 2 Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Chemical Equations Reactions Section 2 Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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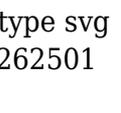
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