

CHM 101: General Chemistry I (2,Units)

Learning Outcomes

At the end of this course, the students should be able to:

1. define atom, molecules and chemical reactions;
2. discuss the Modern electronic theory of atoms;
3. write electronic configurations of elements on the periodic table;
4. justify the trends of atomic radii, ionization energies, electronegativity of the elements based on their position in the periodic table;
5. identify and balance oxidation – reduction equation and solve redox titration problems;
6. illustrate shapes of simple molecules and hybridized orbitals;
7. identify the characteristics of acids, bases and salts, and solve problems based on their quantitative relationship;
8. apply the principles of equilibrium to aqueous systems using LeChatelier's principle to predict the effect of concentration, pressure and temperature changes on equilibrium mixtures;
9. analyse and perform calculations with the thermodynamic functions, enthalpy, entropy and free energy; and
10. determine rates of reactions and its dependence on concentration, time and temperature.

Course Contents

Atoms, molecules, elements and compounds and chemical reactions. Modern electronic theory of atoms. Electronic configuration, periodicity and building up of the periodic table. Hybridization and shapes of simple molecules. Valence Forces and structure of solids. Chemical equations and stoichiometry, chemical bonding and intermolecular forces, kinetic theory of matter. Elementary thermochemistry, rates of reaction, equilibrium and thermodynamics. Acids, bases and salts. Properties of gases. Redox reactions and introduction to electrochemistry. Radioactivity.

Chm 101 Noun Course Material

Joacim Rocklöv



Chm 101 Noun Course Material:

The book delves into Chm 101 Noun Course Material. Chm 101 Noun Course Material is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chm 101 Noun Course Material, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chm 101 Noun Course Material
- Chapter 2: Essential Elements of Chm 101 Noun Course Material
- Chapter 3: Chm 101 Noun Course Material in Everyday Life
- Chapter 4: Chm 101 Noun Course Material in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chm 101 Noun Course Material. The first chapter will explore what Chm 101 Noun Course Material is, why Chm 101 Noun Course Material is vital, and how to effectively learn about Chm 101 Noun Course Material.

3. In chapter 2, this book will delve into the foundational concepts of Chm 101 Noun Course Material. This chapter will elucidate the essential principles that need to be understood to grasp Chm 101 Noun Course Material in its entirety.

4. In chapter 3, this book will examine the practical applications of Chm 101 Noun Course Material in daily life. This chapter will showcase real-world examples of how Chm 101 Noun Course Material can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Chm 101 Noun Course Material in specific contexts. This chapter will explore how Chm 101 Noun Course Material is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Chm 101 Noun Course Material. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chm 101 Noun Course Material.

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Chm 101 Noun Course Material Introduction

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