

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing towards the right. The substance(s) on the left side of the reaction is/are
 - reactant(s)**
 - product(s)
 - solvent
- An ionic chemical equation is represented by adding the
 - charges**
 - structure of the particles
 - number of particles
- An ionic equation shows the activity of aq ions. The other ions are
 - reactants
 - omitted as ions**
 - solvents
 - precipitates
- The activity for precipitation reactions are listed below as they
 - soluble ionic compounds are soluble in the solution (aqueous solution)
 - insoluble ionic compounds
 - soluble ionic compounds
 - insoluble ionic compounds are insoluble in the solution (aqueous solution)**
- The coefficients are chemical equations
 - indicate the number of moles of each substance that react**
 - show the number of grams of each substance that react
 - show the number of molecules of the substances
 - show the relative proportions of each atom
- If x moles of A react, then $2x$ moles of B will be the product, then
 - A will be the reactant
 - B will be the product**
 - B will be the reactant
 - A will be the product

Chapter 8 Test B

**John A. Estrella, Charles Duncan, Sami
Zahran, James L. Haner, Rubin Jen**

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Glencoe Law and Ethics Medical Careers Karen Judson, Sharon Blesie-Hicks, 1998-12

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Holt Science and Technology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001-07

American Physical Education Review, 1924 Includes abstracts of magazine articles and Book reviews

The Open Door Language Series Zenos Edmund Scott, Randolph Thatcher Congdon, Harriet E. Peet, Laura Frazee, 1928

Study Guide for Baumol and Blinder's Economics, Principles and Policy Craig Swan, 1985

The City Record New York (N.Y.), 1915

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Standardized Exercises in United States History Sturgiss Brown Davis, 1921

How to Prepare for and Pass a High School Equivalency Exam I. Edward Friedman, 1966

Understanding Elementary Algebra Arthur Goodman, Steven Kahan, 1998-09 Contains worked out solutions to many

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