

SALT ANALYSIS

1) For $MgSO_4$ - magnesium Sulphate:-

Aim:- To analyze the given salt for an anion and one cation present in it.

Material Required:- Boiling Tube, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

Procedure:-

Sr.	Test	Observation	Inference
i)	Colour	White	
ii)	Odour		
iii)	Flame test	Greenish white	
iv)	Test with H_2SO_4 Sol (Salt + conc. H_2SO_4)	No ppt. form	
v)	Test with H_2SO_4 Sol (Salt + conc. H_2SO_4)	No.	

ii) For $Fe^{2+}Cl^-$, Ferrous chloride :-

Aim:- To analyze the given salt for an anion and one cation present in it.

Material Required:- Boiling Tube, Test Tube, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

Procedure:-

Sr.	Test	Observation	Inference
		Brown reddish	Fe^{2+} may be present
		Colourless	NO_3^- , S^{2-} , CO_3^{2-} are absent
			Ca^{2+} , S^{2-} , SO_4^{2-} , NO_3^- are absent
			Cl^- may be present
			Cl^- confirmed

Class 12th

Chemistry Practical File

Experiment - SALT ANALYSIS

5 most Important Salt

Handwritten PDF Notes

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WJ Hussar



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Table of Contents Chemistry Practical File Class 12 Salt Analysis

1. Understanding the eBook Chemistry Practical File Class 12 Salt Analysis
 - The Rise of Digital Reading Chemistry Practical File Class 12 Salt Analysis
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practical File Class 12 Salt Analysis
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Practical File Class 12 Salt Analysis
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practical File Class 12 Salt Analysis
 - Personalized Recommendations
 - Chemistry Practical File Class 12 Salt Analysis User Reviews and Ratings
 - Chemistry Practical File Class 12 Salt Analysis and Bestseller Lists
5. Accessing Chemistry Practical File Class 12 Salt Analysis Free and Paid eBooks
 - Chemistry Practical File Class 12 Salt Analysis Public Domain eBooks
 - Chemistry Practical File Class 12 Salt Analysis eBook Subscription Services
 - Chemistry Practical File Class 12 Salt Analysis Budget-Friendly Options

6. Navigating Chemistry Practical File Class 12 Salt Analysis eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Practical File Class 12 Salt Analysis Compatibility with Devices
 - Chemistry Practical File Class 12 Salt Analysis Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Practical File Class 12 Salt Analysis
 - Highlighting and Note-Taking Chemistry Practical File Class 12 Salt Analysis
 - Interactive Elements Chemistry Practical File Class 12 Salt Analysis
8. Staying Engaged with Chemistry Practical File Class 12 Salt Analysis
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Practical File Class 12 Salt Analysis
9. Balancing eBooks and Physical Books Chemistry Practical File Class 12 Salt Analysis
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Practical File Class 12 Salt Analysis
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Practical File Class 12 Salt Analysis
 - Setting Reading Goals Chemistry Practical File Class 12 Salt Analysis
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Practical File Class 12 Salt Analysis
 - Fact-Checking eBook Content of Chemistry Practical File Class 12 Salt Analysis
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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