

### balance nuclear reaction equations

aluminum-26 undergoes positron emission  
 potassium-40 undergoes beta decay (electron)  
 uranium-238 undergoes alpha decay



### electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



### NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

- |                             |   |
|-----------------------------|---|
| a. Sulfur hexafluoride      | $\text{SF}_6$   |
| b. Lithium nitride          | $\text{Li}_3\text{N}$   |
| c. Chromium (III) carbonate | $\text{Cr}_2(\text{CO}_3)_3$  |
| d. Tin (II) chloride        | $\text{SnCl}_2$   |
| e. Ammonium acetate         | $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$ ) |
| f. Mercury (I) chloride     | $\text{Hg}_2\text{Cl}_2$  |
| g. Potassium bromate        | $\text{KBrO}_3$   |
| h. Hydrosulfuric acid       | $\text{H}_2\text{S(aq)}$  |
| i. Chloric acid             | $\text{HClO}_3$   |
| j. Sulfurous acid           | $\text{H}_2\text{SO}_3$   |

Name each of the following compounds:

- |                              |                         |
|------------------------------|-------------------------|
| a. $\text{CuSO}_4$           | Copper (II) sulfate     |
| b. $\text{AlF}_3$            | aluminum fluoride       |
| c. HI                        | hydrogen iodide         |
| d. NO                        | nitrogen monoxide       |
| e. $\text{H}_2\text{Se(aq)}$ | hydroselenic acid       |
| f. $\text{HNO}_3$            | nitric acid             |
| g. $\text{HNO}_2$            | nitrous acid            |
| h. $\text{NaHSO}_4$          | sodium hydrogen sulfate |

### Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \rightarrow 20 \rightarrow 10 \rightarrow 5 \rightarrow 2.5 \rightarrow 1.25$$

1            2            3            4            5

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \text{ mg}$$

1            2            3

### Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{r} 63 \times .692 = 43.596 \rightarrow 44 \\ 65 \times .308 = 20.02 \rightarrow 20 \\ \hline 63.62 \rightarrow 64 \end{array}$$

# Chemistry Study Guide Answers Key

**Shasha Hu**



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## **Table of Contents Chemistry Study Guide Answers Key**

1. Understanding the eBook Chemistry Study Guide Answers Key
  - The Rise of Digital Reading Chemistry Study Guide Answers Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Answers Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Study Guide Answers Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Answers Key
  - Personalized Recommendations
  - Chemistry Study Guide Answers Key User Reviews and Ratings
  - Chemistry Study Guide Answers Key and Bestseller Lists
5. Accessing Chemistry Study Guide Answers Key Free and Paid eBooks
  - Chemistry Study Guide Answers Key Public Domain eBooks
  - Chemistry Study Guide Answers Key eBook Subscription Services

- Chemistry Study Guide Answers Key Budget-Friendly Options
- 6. Navigating Chemistry Study Guide Answers Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Study Guide Answers Key Compatibility with Devices
  - Chemistry Study Guide Answers Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Study Guide Answers Key
  - Highlighting and Note-Taking Chemistry Study Guide Answers Key
  - Interactive Elements Chemistry Study Guide Answers Key
- 8. Staying Engaged with Chemistry Study Guide Answers Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Study Guide Answers Key
- 9. Balancing eBooks and Physical Books Chemistry Study Guide Answers Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Study Guide Answers Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Study Guide Answers Key
  - Setting Reading Goals Chemistry Study Guide Answers Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Study Guide Answers Key
  - Fact-Checking eBook Content of Chemistry Study Guide Answers Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web dec 14 2021 the structure of logiks general advanced is verbal 8 minutes 12 questions numerical 8 minutes 8 questions abstract 4 minutes 10 questions the major difference other than structure is test complexity questions at this level are of the same format as those in the individual tests described below

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web the numerical sub test of the logiks general advanced consists of 8 questions with a time limit of 8 minutes these questions assess your numerical reasoning skills and typically consist of a diagram or chart with information and numbers each graph chart will be accompanied by 3 4 questions

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web the cubiks logiks group has five different tests general intermediate general advanced numerical reasoning verbal

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web the cubiks logiks general intermediate test tests 3 major cognitive abilities verbal reasoning numerical reasoning and abstract reasoning all of that with a time limit more so your score will be evaluated in relation to other candidates and only a few are likely to pass seems like a challenge well practicing might just be the way to go

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web similar to logik intermediate logiks advanced tests also assess your verbal numerical and abstract reasoning but with a higher difficulty level the logik advanced tests involve 4 types of tests logiks verbal logiks numerical logiks abstract logiks general advanced logiks verbal

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web fortunately there are preparatory materials available for this exam and you should be able to go into the logiks general advanced test with confidence if you put in the proper amount of time and energy while preparing logiks general advanced test tips practice reading comprehension exercises

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web logiks general intermediate general ability 3 sections verbal numerical and abstract 12 minutes 4 minutes per section

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web apr 17 2023 the logiks general intermediate test is a timed test there are 50 questions in total the test is split into three parts with each part assessing a specific ability numerical verbal reasoning and abstract reasoning