

# Summary: Intermolecular Forces

	Intermolecular Forces	Formed by the attraction between ...	Examples
Strength ↑	Ion-dipole	an ion and a polar molecule	Na <sup>+</sup> and H <sub>2</sub> O
	Hydrogen bond	molecules which have H on N, O, or F atoms	H <sub>2</sub> O and H <sub>2</sub> O; H <sub>2</sub> O and CH <sub>3</sub> CH <sub>2</sub> OH
	Dipole - Dipole	two polar molecules	CH <sub>3</sub> Br and ICl; CH <sub>3</sub> Br and H <sub>2</sub> O
	Ion - Induced dipole	an ion and a nonpolar	Fe <sup>2+</sup> and O <sub>2</sub>
	Dipole - Induced dipole	a polar molecule and a nonpolar molecule	HCl and Cl <sub>2</sub>
	London (dispersion) forces	two nonpolar molecules	CH <sub>4</sub> and CH <sub>4</sub> ; F <sub>2</sub> and F <sub>2</sub> ; CH <sub>4</sub> and F <sub>2</sub>

# Intermolecular Forces And Strengths Pogil Answers

**AW Rasmussen**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

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