Student Exploration: Stoichiometry

Vocabulary: balanced equation, coefficient, conservation of mass, limiting reagent, mole, molecular mass, molecule, reactant, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A chef has a recipe that calls for 2 eggs and 1 cup of flour. She wants to make 4 batches of cookies. How many eggs and how much flour will she need?

2. A recipe for 1 batch of cookies calls for 1 cup of flour and 1 egg. She has 4 cups of flour and 3 eggs. How many batches of cookies can she make?

Gizmo Warm-up

Chemical reactions are described by balanced equations. Balanced equations show the number of molecules of each substance involved in the reaction. The number of molecules of each substance is indicated by a coefficient. For example, the balanced equation for the reaction between hydrogen and oxygen to form water is:

$$2H_2 + O_2 \rightarrow 2H_2O$$

1. Count up the **molecules** in the equation above. How many molecules of each substance are involved in the reaction? How many molecules are produced?

2. In a **balanced equation**, the number of each kind of atom is the same on both sides of the equation. Count the number of each kind of atom in the equation above. Complete the table below.

Element	Atoms of	Coefficient	Subscript	Total
Hydrogen	Atoms of	Coefficient =	Subscript =	Total =
Oxygen	Atoms of	Coefficient =	Subscript =	Total =

Is the equation above balanced? Why, or why not?

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