

- 4 Chlorine has two isotopes, ${}^{35}_{17}\text{Cl}$ and ${}^{37}_{17}\text{Cl}$.
- (a) What do you understand by the term *isotope*? [1]
- (b) Write down the number of neutrons in each of these isotopes of chlorine. [2]
- (c) Draw and label a diagram of an atom of chlorine to show:
- the nucleus
 - the electron shells, with the correct number of electrons in each shell. [3]
- (d) A chloride ion has eight electrons in its outer shell. What is the importance of this electronic configuration? [1]

(Paper 3)

Chapter 2 Section Review Questions

LM Reder

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web below are some important points to get clear mole concept definition of mole chemistry and mole units mole can be define as the atomic mass molecular mass ionic mass or formula mass of a substance expressed in grams is called mole of

that substance

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web libretexts the mole concept is central to any chemistry calculation based on experimental results the mole is how we relate the unbelievably small atoms and molecules that make something up to the measurable properties such as mass which we may observe in a laboratory setting

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web oct 10 2023 mole standard unit $6.02214076 \times 10^{23}$ in chemistry for measuring large quantities of very small entities such as atoms molecules or other specified particles the number of units in a mole also bears the name avogadro s number or avogadro s constant in honor of the italian physicist amedeo avogadro

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web the mole was made the seventh si base unit in 1971 by the 14th cgpm 2019 redefinition of si base units before the 2019 redefinition of the si base units the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 12 grams of carbon 12 the most common isotope of carbon