

Thermodynamic Equilibrium - 1 (Multiple choice)

Multiple choice

Identify the correct statement about each condition for thermodynamic equilibrium between two systems.

1. The temperatures of two systems are different, but the (thermodynamic) entropies of the systems are equal.
- True
- False
- All of the above.
2. The two thermodynamic states have the same pressure and temperature, but not the same volume.
- True
- False
3. The two systems are both expanding, but neither is doing any work, and neither is exchanging heat with the surroundings.
- True
- False
- All of the above.
4. The two systems are both expanding, but neither is doing any work, and neither is exchanging heat with the surroundings.
- True
- False
- All of the above.
5. The two thermodynamic states are at the same pressure and temperature, but the systems are not in contact.
- Equilibrium is not possible.
- Equilibrium is not possible.
- Equilibrium is possible.
- Equilibrium is not possible.
6. The two thermodynamic states are at the same pressure and temperature, but the systems are not in contact.
- True
- False
- All of the above.
7. The two thermodynamic states are at the same pressure and temperature, but the systems are not in contact.
- True
- False
- All of the above.
8. The two thermodynamic states are at the same pressure and temperature, but the systems are not in contact.
- True
- False
- All of the above.
9. The two thermodynamic states are at the same pressure and temperature, but the systems are not in contact.
- Equilibrium is not possible.
- Equilibrium is not possible.
- Equilibrium is possible.
- Equilibrium is not possible.

Chapter Test Thermal Energy Answers

DJ Losen



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