

# 2. Chemistry of Chapter 2 (Exercises 1-10)



## Exercise 1

Which of the following is not an ionic compound?

- a. sodium chloride
- b. sodium acetate
- c. sodium hydroxide
- d. sodium cyanide
- e. sodium nitrate

1  
2  
3  
4  
5  
6

- 1. empirical formula/molar mass of sodium acetate
- 2. calculate empirical formula of sodium acetate and compare with molar formula of the compound
- 3. molar mass of sodium acetate
- 4. calculate the percent composition of sodium acetate
- 5. comparison of the percent composition of sodium acetate
- 6. identification of the compound

Which of the following is not an ionic compound?

- a. sodium hydroxide
- b. sodium acetate
- c. sodium cyanide
- d. sodium nitrate
- e. sodium chloride

7  
8  
9  
10  
11  
12

- 7. sodium hydroxide
- 8. sodium acetate
- 9. sodium cyanide
- 10. sodium nitrate
- 11. sodium chloride
- 12. comparison of the percent composition of sodium acetate

## Exercise 2

Which of the following is not an ionic compound?

13  
14  
15  
16  
17

- 13. sodium chloride
- 14. sodium acetate
- 15. sodium cyanide
- 16. sodium nitrate
- 17. sodium hydroxide

18  
19  
20  
21  
22

- 18. sodium
- 19. acetate
- 20. cyanide
- 21. nitrate
- 22. hydroxide

# Modern Chemistry Chapter 5 Test Answers

**S Baum**



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