

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ A. CH_3COOH
- _____ B. Lewis Acids
- _____ C. Carbonyl Compounds
- _____ D. Alcohols
- _____ E. Metal Ions
- _____ F. Simple Molecules
- _____ G. Molecules
- _____ H. Lewis Acids
- _____ I. Hydrogen Ions
- _____ J. None

- _____ A. bonded to electronegative atoms that are more electronegative than carbon
- _____ B. single group of negatively charged atoms
- _____ C. the energy released when one mole of an ionic crystalline compound dissolves in water
- _____ D. coordination bonds are part of structure in which both atoms have lone pairs
- _____ E. the energy released in bonds is identical to the energy released in bonds
- _____ F. bonded group of atoms that are held together by covalent bonds, results from the sharing of electron pairs between atoms
- _____ G. chemical bonding that results from the electrical attraction between positive and negative charges
- _____ H. mutual electrical attraction between protons and electrons, and between atoms, that holds the atoms together

Multiple Choice

2. The charge on the bond is _____

- _____ A. always positive
- _____ B. always negative
- _____ C. either positive or negative
- _____ D. zero

3. In many compounds, atoms of more electronegative elements form bonds with the atoms of less electronegative elements. The electronegativity difference between the two atoms is _____

- _____ A. 0
- _____ B. 1
- _____ C. 2
- _____ D. 3

4. Which of the following is not a Lewis acid? _____

- _____ A. CH_3COOH
- _____ B. Lewis Acids
- _____ C. Carbonyl Compounds
- _____ D. Alcohols
- _____ E. Metal Ions
- _____ F. Simple Molecules
- _____ G. Molecules
- _____ H. Lewis Acids
- _____ I. Hydrogen Ions
- _____ J. None

Modern Chemistry Chapter 6 Test Answers

B Lingard



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