Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
   a) White
   b) Cream
   c) Yellow
   d) Green

2. What colour do sodium ions go in a flame test?
   a) Red
   b) Yellow
   c) Brick-red
   d) Lilac

3. Which two solutions are added to test for the presence of halide ions?
   a) Silver chloride and nitric acid
   b) Barium chloride and hydrochloric acid
   c) Silver nitrate and nitric acid
   d) Barium chloride and sulphuric acid

4. What is the test for oxygen gas?
   a) Lit splint goes out with a squeaky pop
   b) Damp blue litmus paper goes red
   c) Damp red litmus paper goes blue
   d) Glowing splint relights

5. What colour precipitate to iodide ions form?
   a) White
   b) Cream
   c) Yellow
   d) Green

6. Which ion produces a brick-red colour in a flame test?
   a) Sodium
   b) Calcium
   c) Copper
   d) Lithium

7. What is the test for carbon dioxide gas?
   a) Limewater turns cloudy
   b) Damp blue litmus turns red
   c) A glowing splint relights
   d) A lit splint goes out with a squeaky pop

8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
   a) White
   b) Blue
   c) Green
   d) Red/brown

9. What is the test for chlorine gas?
   a) Damp blue litmus paper turns red and is then bleached white
   b) Red litmus [] turns blue
   c) A glowing splint relights
   d) Limewater will turn cloudy

10. Copper (II) ions form what colour precipitate test?
    a) Blue
    b) Green
    c) Red/brown
    d) Cream

11. Which two solutions are added to test for the presence of sulphate ions?
    a) Silver chloride and nitric acid
    b) Barium chloride and sulphuric acid
    c) Silver nitrate and nitric acid
    d) Barium chloride and hydrochloric acid

12. What colour do potassium ions go in a flame test?
    a) Lilac
    b) Red
    c) Yellow
    d) Brick-red

13. What is the test for hydrogen gas?
    a) Lit splint goes out with a squeaky pop
    b) Glowing splint relights
    c) Damp blue litmus paper turns red and then is bleached white
    d) Limewater goes cloudy

14. Which ion produces a red colour in a flame test?
    a) Sodium
    b) Calcium
    c) Copper
    d) Lithium

15. What colour precipitate to bromide ions form?
    a) White
    b) Cream
    c) Yellow
    d) Green

16. What needs to be added to test for the presence of carbonate ions?
    a) Silver nitrate
    b) Universal indicator
    c) Litmus paper
    d) Dilute acid

17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
    a) Oxygen
    b) Ammonia
    c) Carbon dioxide
    d) Hydrogen

18. What colour precipitate does barium sulphate form?
    a) White
    e) Green
    f) Yellow
    g) Cream

19. What is needed to carry out a flame test?
    a) Splint
    b) Glass rod
    c) Spatula
    d) Nichrome/platinum wire

Your score out of 19=
Percentage =

Chemistry Unit 2 Test Answers

Michael Brown

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