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ASHOK K. GUPTA

3rd EDITION



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so calculating enthalpy change of reaction in $2 \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_6) - 6 \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_4) = 6(-84.7) - 6(52.3) = -508.2 - 313.8 = -822 \text{ kJ}$

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