

Name _____ Period _____

Acid-Base Solutions – PhET Simulator

Instructions: Open the Acid-Base Solutions simulator on the PhET website/app. Choose "Introduction."

- **pH Test Strips** change color when they come in contact with a solution.
- The color of the strip is compared to a **pH Color Key** to determine the pH.



- 1) Select the **pH Test Strip** from the **Tools** menu. Dip it into water. What is the pH? 7
- 2) Use pH test strips to test the **Strong Acid**, **Weak Acid**, **Strong Base**, and **Weak Base**. Draw the symbol for each solution in the appropriate location on the **pH Color Key** below:



Draw each symbol under the correct pH

SA	WA	WB	SB
Strong Acid	Weak Acid	Strong Base	Weak Base

- When acids and bases dissolve in water, their ions **dissociate** (separate).
- Acids dissociate by releasing **H⁺ ions**. These **H⁺ ions** bond to **H₂O** to form **H₃O⁺**.

- 3) Choose the **Strong Acid**. Complete the chemical formula for dissociation of a strong acid:



Instructions: Under "Views," choose the **Graph** option.

- 4) What is the concentration of **ions** (A⁻ and H₃O⁺) in the **Strong Acid**? 1.00×10^{-2}
- 5) What is the concentration of **ions** (A⁻ and H₃O⁺) in the **Weak Acid**? 3.16×10^{-5}
- 6) Which type of acid (**Strong** or **Weak**) produces more **ions**? **Strong Acid**



DIGITAL LAB

Acid And Bases Ph Phet Lab Answers

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Using Indicators | Grade 6-8 Physical Science Baby Professor,2024-04-15 Dive into the intriguing world of acids bases and the pH scale with Oh pH Understanding H OH and pH of Solutions and Using Indicators a must have for grades 6 8 in the US STEM curriculum This book simplifies complex concepts into engaging lessons including hands on experiments to distinguish between acids and bases using everyday items Essential for enhancing science education it offers a foundational understanding of chemical reactions and their impacts Perfect for teachers homeschooling parents and librarians it s a call to explore the fascinating chemistry surrounding us Add it to your educational resources today *Acid-base Equilibria* Edward Jasper King,1965 Band 4 Acid-Base Case Studies Ira Kurtz,2004 The general aim of this book is to present a practical case orientated approach to the analysis of acid base problems in the clinical setting Medical students residents fellows and attendings will find the book a required addition to their medical library **High-yield Acid-base** J. Craig Longenecker,2007 The goal of this book is to provide a bridge between the acid base physiology taught in the classroom and the evaluation of the patient on the wards This book will enable the reader to develop a practical and reasoned approach to the patient with an acid base disorder Additional resources at the back of the book include an abbreviation list to familiarize readers with common terms associated with acid base pathophysiology and a comprehensive list of suggested readings Acid Base Indicators I. M. Kolthoff,2007-03-01 Many of the earliest books particularly those dating back to the 1900s and before are now extremely scarce and increasingly expensive We are republishing these classic works in affordable high quality modern editions using the original text and artwork **Acid-base Balance; Chemistry, Physiology, Pathophysiology** Arthur Gorman Hills,1973

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