

Chemical Equations and Reactions

QUIZ
CH #8
SECTION #1

Section Quiz: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- B 1. Which of the following is true regarding the law of conservation of mass?
- It dictates that the number of molecules on each side of a chemical equation must be the same.
 - It dictates that the number of atoms of each element must be the same on both sides of a chemical equation.
 - It states that the mass of the reactants must remain constant in order for a chemical reaction to proceed.
 - It does not apply to chemical reactions.
- D 2. Which of the following may indicate that a chemical reaction has occurred?
- release of energy as light
 - a color change
 - gas bubble formation
 - All of the above
- D 3. Which of the following is a product in the reaction described by the word equation below?
- iron + copper(II) sulfate \rightarrow iron(II) sulfate + copper
- iron(II) sulfate
 - copper
 - iron
 - Both (a) and (b)
- C 4. What does CH_4 represent in the balanced chemical equation shown below?
- $$\text{C} + 2\text{H}_2 \rightarrow \text{CH}_4$$
- \uparrow
 -1
 -1
- one atom of CH_4
 - five atoms of CH_4
 - one molecule of CH_4
 - 1 g of CH_4
- C 5. Which of the following is not true regarding a properly written chemical equation?
- It must be balanced.
 - It must contain proper chemical formulas.
 - It must show the formation of a precipitate or a gas.
 - It should represent known data.

Chapter 8 Chemical Equations And Reactions Test Answers

Arshad Iqbal



Chapter 8 Chemical Equations And Reactions Test Answers:

Chapter 8 Chemical Equations And Reactions Test Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chapter 8 Chemical Equations And Reactions Test Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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