

Thermodynamic Equilibrium - 1 (Application of Equilibrium)

Thermodynamic Equilibrium

Thermodynamic Equilibrium: A system is in thermodynamic equilibrium if its macroscopic properties do not change with time.

1. In a closed system at constant volume and pressure, which of the following is not a state function?
- A. Entropy
 - B. Heat
 - C. pH of the solution
2. For the thermodynamic system shown below, which process is not a state function?
- A. Entropy
 - B. Heat
 - C. pH
3. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Entropy
 - B. Heat
 - C. Enthalpy
 - D. None of the above
4. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Heat
 - B. Enthalpy
 - C. Entropy
 - D. None of the above
5. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Enthalpy
 - B. Heat
 - C. Entropy
 - D. None of the above
6. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Entropy
 - B. Heat
 - C. pH
7. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Heat
 - B. Enthalpy
 - C. Entropy
 - D. None of the above
8. Which of the following is not a state function? (Consider the following: ΔH is the change in enthalpy, ΔS is the change in entropy, ΔG is the change in Gibbs free energy, and ΔU is the change in internal energy.)
- A. Enthalpy
 - B. Heat
 - C. Entropy
 - D. None of the above

Chapter Test Thermal Energy Answers 16

Lauren Gardner



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