

CH-101 2022 Assignment 10 Cases

Name: _____

Number: _____

10. Atmospheric pressure at elevations of 1000 ft average about 0.70 atm. Consider this pressure as well-points of mercury, tubes of mercury, and provide your answer with respect to atmospheric pressure at 1000 feet mean sea level (Patmos Island)?

Answer: At Patmos, atmospheric pressure is lower than atmospheric pressure at sea level. The pressure at 1000 ft is approximately 0.70 atm, which is lower than the atmospheric pressure at sea level (1.00 atm).

11. Suppose a tube of mercury is 1.00 m long. Suppose the pressure is 1.00 atm. What would happen to the volume of the tube? What if the end of the tube was in a liquid that was in the same pressure as the atmosphere? What if the end of the tube was in a liquid that was in a higher pressure? What if the end of the tube was in a liquid that was in a lower pressure?

12. A volume of gas is confined in a cylinder with a freely moving piston, and heat is added to the gas. The pressure in the cylinder is held constant. The volume of the gas increases. The pressure in the cylinder is held constant. The volume of the gas increases. The pressure in the cylinder is held constant. The volume of the gas increases. The pressure in the cylinder is held constant. The volume of the gas increases.

- (a) The pressure will increase to half that the volume will double.
- (b) The pressure will increase to half that the volume will double.
- (c) The pressure will increase to half that the volume will double that number.
- (d) The pressure will increase to half that the volume will double.

13. Use Charles's law to compare the volume of a gas at 100°C and 200°C. The volume of the gas is 1.00 L at 100°C. What is the volume of the gas at 200°C?

Temperature (°C)	Volume (L)	Temperature (°C)	Volume (L)
100	1.00	200	1.00
100	1.00	200	1.00
100	1.00	200	1.00
100	1.00	200	1.00

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