

CHEMICAL BONDING POGIL

Activity 3 – What's in a name?

Objective:

- Identify some simple rules about nomenclature (naming).

The Model:

Examine the table below, and answer the following questions.

Table 2

Cation	Anion	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Ca^{2+}	O^{2-}	CaO	calcium oxide
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Li^+	S^{2-}	Li_2S	lithium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Reviewing the Model

- Are ALL cations positive ions or negative ions? *positive*
- Are ALL anions positive ions or negative ions? *negative*
- What is the name of the compound formed by the combination of Li^+ and S^{2-} ions? *Lithium sulfide*

Exploring the Model

- When the name of an ionic compound is given, which ion is stated first?
Cation
- Compare the first part of the compound name to the name of the element from the periodic table. How does the name of the cation correspond to the name of the element?
- Compare the second part of the compound name to the name of the element from the periodic table. How does the name of the anion correspond to the name of the element?
- From what part of the periodic table do the cations in the Model come (metals or nonmetals)?
Alkali and alkaline earth
- From what part of the periodic table do the anions in the Model come?
4

Chemical Bonding Pogil Answers Key

KJ Lindholm-Leary



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