

CHEMISTRY MULTIPLE CHOICE QUESTIONS

A. Atoms, molecules and stoichiometry

2002 -2019



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C.

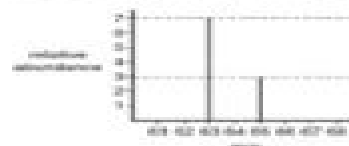
As a simplification, an adult human can be considered to have a daily diet of 1.80 kg of carbohydrate (empirical formula $C_6H_{12}O_6$).

Which mass of carbon dioxide does a person produce each day if all the carbohydrate eaten is digested and oxidised?

- A. 0.907 kg B. 0.900 kg C. 1.30 kg D. 2.54 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A. 63.2 B. 63.5 C. 63.8 D. 64.0

(2002 NA/10)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 MC/31)

A mixture of 1.00 g of methane and 1.00 g of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula CH_4O ?

- methanol
- ethanoic acid
- methyl methanolate

(2002 ON/31)

What is the number of molecules in 500 cm³ of oxygen under room conditions?

- A. 1.25×10^{23} B. 1.25×10^{24} C. 3.0×10^{23} D. 3.0×10^{24}

(2008 MC/12)

In the preparation of soft margarine, glycerol trioleate reacts



is fully hydrogenated so that, on average, one of its side-chains is converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2(\text{CH}_2)_7\text{CO}_2$ moiety and two side-chains are converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2(\text{CH}_2)_7\text{CO}_2$ moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A. 4 B. 5 C. 6 D. 8

(2008 MC/22)

Use of the Data Booklet is relevant to this question.

Additional classes are about very small amounts of carbon which dissolve in 1000 g of

Chemistry Multiple Choice Questions With Answers

Henry Lester Gerry



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covers topics of chemical to electrical energy applications of electrolysis reactions conductors and non conductors dry cells electrical devices circuit symbols electrolytes non electrolytes organic solvents polarization and valence electrons The chapter Electricity and Chemicals MCQs covers topics of chemical to electrical energy dry cells electrolyte non electrolyte and polarization The chapter Elements Compounds and Mixtures MCQs covers topics of elements compounds mixtures molecules atoms and symbols for elements

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Examination Questions with Answers Daniel B. Murphy, 1965

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Multiple Choice Questions With Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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