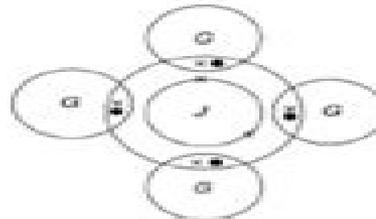


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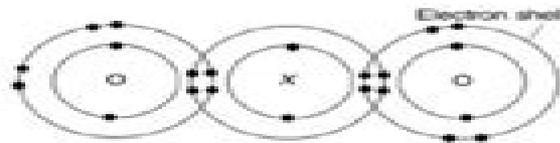
Paper 1

- 1) The diagram below shows the electron arrangement for the compound that is formed between atom *J* and atom *G*.



Which of the following statements is true about the compound?

- A The compound conducts electricity
  - B The compound is an ionic compound
  - C The melting point of the compound is low
  - D The atoms within the compound are attracted together by electrostatic force
- 2) What is the electron arrangement of atom *X*?



- A 2.4
  - B 2.6
  - C 2.8.6
  - D 2.8.4
- 3) The table below shows the information regarding two types of particles.

| Particle | Number of protons | Electron configuration |
|----------|-------------------|------------------------|
| <i>P</i> | 8                 | 2.8                    |
| <i>Q</i> | 9                 | 2.8                    |

Both particles *P* and *Q* are

- A halide ions
- B noble gases
- C cations
- D anions

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