

Chemical Analysis – What can you Remember?

- What colour precipitate do chloride ions produce?
 - White
 - Cream
 - Yellow
 - Green
- What colour do sodium ions go in a flame test?
 - Red
 - Yellow
 - Brick-red
 - Lilac
- Which two solutions are added to test for the presence of halide ions?
 - Silver chloride and nitric acid
 - Barium chloride and hydrochloric acid
 - Silver nitrate and nitric acid
 - Barium chloride and sulphuric acid
- What is the test for oxygen gas?
 - Lit splint goes out with a squeaky pop
 - Damp blue litmus paper goes red
 - Damp red litmus paper goes blue
 - Glowing splint relights
- What colour precipitate do iodide ions form?
 - White
 - Cream
 - Yellow
 - Green
- Which ion produces a brick-red colour in a flame test?
 - Sodium
 - Calcium
 - Copper
 - Lithium
- What is the test for carbon dioxide gas?
 - Lime water turns cloudy
 - Damp blue litmus turns red
 - A glowing splint relights
 - A lit splint goes out with a squeaky pop
- When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - White
 - Blue
 - Green
 - Red/brown
- What is the test for chlorine gas?
 - Damp blue litmus paper turns red and is then bleached white
 - Red litmus turns blue
 - A glowing splint relights
 - Lime water will turn cloudy
- Copper (II) ions form what colour precipitate test?
 - Blue
 - Green
 - Red/brown
 - Cream
- Which two solutions are added to test for the presence of sulphate ions?
 - Silver chloride and nitric acid
 - Barium chloride and sulphuric acid
 - Silver nitrate and nitric acid
 - Barium chloride and hydrochloric acid
- What colour do potassium ions go in a flame test?
 - Lilac
 - Red
 - Yellow
 - Brick-red
- What is the test for hydrogen gas?
 - Lit splint goes out with a squeaky pop
 - Glowing splint relights
 - Damp blue litmus paper turns red and then is bleached white
 - Lime water goes cloudy
- Which ion produces a red colour in a flame test?
 - Sodium
 - Calcium
 - Copper
 - Lithium
- What colour precipitate do bromide ions form?
 - White
 - Cream
 - Yellow
 - Green
- What needs to be added to test for the presence of carbonate ions?
 - Silver nitrate
 - Universal indicator
 - Litmus paper
 - Dilute acid
- If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - Oxygen
 - Ammonia
 - Carbon dioxide
 - Hydrogen
- What colour precipitate does barium sulphate form?
 - White
 - Green
 - Yellow
 - Cream
- What is needed to carry out a flame test?
 - Splint
 - Glass rod
 - Spatula
 - Nichrome/platinum wire

Your score out of 19 =
Percentage =

Chemistry Unit 2 Test Answers

J Spring



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