

# COMBINED GAS LAW

In practical terms, it is often difficult to hold any of the variables constant. When there is a change in pressure, volume and temperature, the combined gas law is used.

$$\frac{P_1 \times V_1}{T_1} = \frac{P_2 \times V_2}{T_2} \quad \text{or} \quad P_1 V_1 T_2 = P_2 V_2 T_1$$

Complete the following chart.

	$P_1$	$V_1$	$T_1$	$P_2$	$V_2$	$T_2$
1	1.5 atm	3.0 L	20° C	2.5 atm	1.2 L	30° C
2	720 torr	256 mL	25° C	368.64 torr	250 mL	50° C
3	600 mmHg	2.5 L	22° C	760 mmHg	1.8 L	24.1225° C
4	25000 atm	750 mL	0.0° C	2.0 atm	500 mL	25° C
5	95 kPa	4.0 L	315.76° C	101 kPa	6.0 L	471 K or 198° C
6	650. torr	467.3 mL	100° C	900. torr	225 mL	150° C
7	850 mmHg	1.5 L	15° C	255 mmHg	2.5 L	30° C
8	125 kPa	125 mL	48° C	100 kPa	100 mL	75° C

# Combined Gas Law Problems Chemfiesta Answer Key

**ML Yell**



**Combined Gas Law Problems Chemfiesta Answer Key:**

The Enigmatic Realm of **Combined Gas Law Problems Chemfiesta Answer Key**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Combined Gas Law Problems Chemfiesta Answer Key** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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