

Ideal gas law - problems and solutions

1. Ideal gases in a closed container initially have volume V and temperature T . The final temperature is $5/4T$ and the final pressure is $2P$. What is the final volume of gases.

Known :

Initial volume (V_1) = V

Initial temperature (T_1) = T

Final temperature (T_2) = $5/4 T$

Initial pressure (P_1) = P

Final pressure (P_2) = $2P$

Wanted : Final volume (V_2)

Solution :

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$\frac{P V}{T} = \frac{(2P) V_2}{5/4 T}$$

$$\frac{V}{1} = \frac{2 V_2}{5/4}$$

$$5/4 V = 2 V_2$$

$$V_2 = \frac{5V}{4} \cdot \frac{2}{1}$$

$$V_2 = \frac{5V}{4} \times \frac{1}{2}$$

$$V_2 = \frac{5V}{8}$$

2. Determine the volume of 2.00 moles of gases (ideal gas) at STP. STP = Standard Temperature and Pressure.

Known :

Moles of gas (n) = 2 moles

Standard temperature (T) = $0^\circ\text{C} = 0 + 273 = 273$ Kelvin

Standard pressure (P) = 1 atm = 1.013×10^5 Pa

Universal gas constant (R) = 8.315 Joule/mole.Kelvin

Wanted : Volume of gases (V)

Solution :

Read more :

<https://physics.gurumuda.net/ideal-gas-law-problems-and-solutions.htm>

Gas Law Problems With Solutions

Ensheng Dong



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Journal of Chemical Education, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers

Questions and Answers on Federal Tax Laws, Problems Affecting Individuals, Partnerships and Corporations, January, 1924 Irving Bank-Columbia Trust Company, 1924

Math Concepts for Food Engineering Richard W. Hartel, D.B. Hyslop, Robin K. Connelly, T.A. Howell Jr., 2008-03-17 A Supplement for Food Science Engineering Students Who Need to Improve Their Mathematical Skills A remedial textbook for understanding mathematical theories and formulas Math Concepts for Food Engineering Second Edition helps students improve their mathematical skills so that they can succeed in food engineering course

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change

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