

## General Chemistry Questions

### Electronic Structure and Periodic Table

1. What value or values of  $m_l$  are allowable for an orbital with  $l = 2$ ?
  - a. 0
  - b. 2
  - c. -1
  - d. none of the above
  - e. all of the above
2. According to Bohr Theory, which of the following transitions in the hydrogen atom will give rise to the **least energetic** photon?  
Use the equation:  $E_n = (-2.18 \times 10^{-18} \text{ J})(1/n^2)$ 
  - a.  $n = 5$  to  $n = 3$
  - b.  $n = 6$  to  $n = 1$
  - c.  $n = 4$  to  $n = 3$
  - d.  $n = 5$  to  $n = 4$
  - e.  $n = 6$  to  $n = 5$
3. Consider a  $3d_{zz}$  orbital. Which of the following statements is **incorrect**?
  - a. The  $xz$  plane is a nodal surface.
  - b. The  $xz$  plane divides the electron probability distribution into two identical mirror-image halves.
  - c. The  $xy$  plane divides the electron probability distribution into two identical mirror-image halves.
  - d. The  $yz$  plane divides the electron probability distribution into two identical mirror-image halves.
  - e. The nucleus is located at a node.
4. The electronic configuration of the element whose atomic number is 26 is:
  - a.  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^0 3d^6$
  - b.  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^2$
  - c.  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
  - d.  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6 4p^2$
  - e. none of the above
5. Which of the following has the largest radius?
  - a. F
  - b. N
  - c. C
  - d. O
  - e. Ne

# General Chemistry Questions Answers

**AN Whitehead**



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## **Table of Contents General Chemistry Questions Answers**

1. Understanding the eBook General Chemistry Questions Answers
  - The Rise of Digital Reading General Chemistry Questions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Questions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an General Chemistry Questions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Questions Answers
  - Personalized Recommendations
  - General Chemistry Questions Answers User Reviews and Ratings
  - General Chemistry Questions Answers and Bestseller Lists

5. Accessing General Chemistry Questions Answers Free and Paid eBooks
  - General Chemistry Questions Answers Public Domain eBooks
  - General Chemistry Questions Answers eBook Subscription Services
  - General Chemistry Questions Answers Budget-Friendly Options
6. Navigating General Chemistry Questions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - General Chemistry Questions Answers Compatibility with Devices
  - General Chemistry Questions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of General Chemistry Questions Answers
  - Highlighting and Note-Taking General Chemistry Questions Answers
  - Interactive Elements General Chemistry Questions Answers
8. Staying Engaged with General Chemistry Questions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers General Chemistry Questions Answers
9. Balancing eBooks and Physical Books General Chemistry Questions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection General Chemistry Questions Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine General Chemistry Questions Answers
  - Setting Reading Goals General Chemistry Questions Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of General Chemistry Questions Answers
  - Fact-Checking eBook Content of General Chemistry Questions Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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