

# **Chapter 3**

## **The Business Research Process: An Overview**

# Research Methods For Business Uma Sekaran Chapter 4 Ppt

**LP Steffe**



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web limiting reactants chemistry socratica limiting reactants what is a limiting reactant it s not immediately obvious when you look at a chemical equation for a reaction when you see a chemical equation in a textbook it looks like the reactants are completely used up and all made into product

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web jun 2 2020 the reactant that is consumed first and limits the amount of product s that can be obtained is the limiting reactant to identify the limiting reactant calculate the number of moles of each reactant present and compare this ratio to the mole ratio of the reactants in the balanced chemical equation

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web jun 30 2023 the limiting reagent is the one that is totally consumed it limits the reaction from continuing because there is none left to react with the in excess reactant there are two ways to determine the limiting reagent

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web a which chemical is the limiting reactant b how many grams of zns will be formed c how many grams of the excess reactant will remain after the reaction is over 3 which element is in excess when 3 00 grams of mg is ignited in 2 20 grams of

pure oxygen what mass is in excess what mass of  $\text{MgO}$  is formed 4

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web reactant present in an amount lower than required by the reaction stoichiometry thus limiting the amount of product generated molecular equation chemical equation in which all reactants and products are represented as neutral substances

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web the key to recognizing which reactant is the limiting reagent is based on a mole mass or mass mass calculation

whichever reactant gives the lesser amount of product is the limiting reagent what we need to do is determine an amount of one product either moles or mass assuming all of each reactant reacts

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