

# Chemistry Questions and Answers

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1. 15. Why do we store silver chloride in dark coloured bottle?

Silver chloride decomposes in presence of light, forming silver and chlorine gas. Storing in dark bottles prevents exposure to light, thus preserving it. This is an example of photodecomposition. Protecting it from light maintains the chemical's stability, avoiding undesired reactions that can alter its composition or effectiveness in experiments or applications.

2. 16. A metal when exposed to air forms black coating over it. Name the process and the metal compound formed by the process.

The black coating is due to corrosion or oxidation. The process is tarnishing. When a metal like silver is exposed to air, it reacts with sulfur compounds forming black silver sulfide ( $\text{Ag}_2\text{S}$ ). This is a chemical reaction involving atmospheric gases, which forms a layer preventing further corrosion temporarily, but eventually causes damage.

3. 17. a) Write two observations when lead nitrate is heated in a test tube.  
b) Name the type of reaction.

a) Lead nitrate melts and emits a brown gas.

b) It is a thermal decomposition reaction. On heating, lead nitrate breaks down to form lead oxide, nitrogen dioxide, and oxygen. This reaction involves a visible change and emission of gas, indicating a chemical change due to heat-induced decomposition of the compound.

4. 18. Potato chips manufacturers fill the packet of chips with nitrogen gas. Give reason.

Nitrogen is an inert gas that doesn't react easily. It is filled in chip packets to displace oxygen, which can cause rancidity in oils. Nitrogen prevents spoilage and keeps the chips fresh for longer. It also maintains the shape of the packet, acting as a cushion during transportation and storage.

5. 19. When barium hydroxide is mixed with ammonium chloride, a fall in temperature is observed. Why?

The reaction between barium hydroxide and ammonium chloride is endothermic. It absorbs heat, causing a drop in temperature. This indicates energy is consumed to break bonds and form new ones. It is a good example of how chemical reactions can affect temperature, showing the transfer and transformation of energy in reactions.

6. 20. Write the chemical equation of the reaction with an example each in which the following has taken place.  
a) change in temperature.

# Industrial Chemistry Question Answer

**S Ben Porath**



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