

B-2-22. $\ddot{x} + 4x = 0, \quad x(0) = 5, \quad \dot{x}(0) = 0$

The Laplace transform of the given differential equation is

$$[s^2X(s) - sx(0) - \dot{x}(0)] + 4X(s) = 0$$

Substitution of the initial conditions into this last equation gives

$$(s^2 + 4)X(s) = 5s$$

Solving for $X(s)$, we obtain

$$X(s) = \frac{5s}{s^2 + 4}$$

The inverse Laplace transform of $X(s)$ is

$$x(t) = 5 \cos 2t$$

This is the solution of the given differential equation.

B-2-23. $\ddot{x} + \omega_n^2 x = t, \quad x(0) = 0, \quad \dot{x}(0) = 0$

The Laplace transform of this differential equation is

$$s^2X(s) + \omega_n^2 X(s) = \frac{1}{s^2}$$

Solving this equation for $X(s)$, we obtain

$$X(s) = \frac{1}{s^2(s^2 + \omega_n^2)} = \left(\frac{1}{s^2} - \frac{1}{s^2 + \omega_n^2} \right) \frac{1}{\omega_n^2}$$

The inverse Laplace transform of $X(s)$ is

$$x(t) = \frac{1}{\omega_n^2} \left(t - \frac{1}{\omega_n} \sin \omega_n t \right)$$

This is the solution of the given differential equation.

B-2-24. $2\ddot{x} + 2\dot{x} + x = 1, \quad x(0) = 0, \quad \dot{x}(0) = 2$

The Laplace transform of this differential equation is

$$2[s^2X(s) - sx(0) - \dot{x}(0)] + 2[sX(s) - x(0)] + X(s) = \frac{1}{s}$$

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**Bernard Sandler Meyer, Donald Benton
Anderson**

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Global Crises, Global Solutions Bjørn Lomborg, 2009-07-09 In this book leading economists evaluate how the world can best spend money to combat the world's biggest problems

Thermodynamics of Solutions Eli Ruckenstein, Ivan L. Shulgin, 2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade. Even though they involve different topics and different systems, they have something in common which can be considered as the signature of the present book. First, these papers are concerned with difficult or very nonideal systems, i.e., systems with very strong interactions, e.g., hydrogen bonding between components or systems with large differences in the partial molar volumes of the components, e.g., the aqueous solutions of proteins or systems that are far from normal conditions, e.g., critical or near-critical mixtures. Second, the conventional thermodynamic methods are not sufficient for the accurate treatment of these mixtures. Last but not least, these systems are of interest for the pharmaceutical, biomedical, and related industries. In order to meet the thermodynamic challenges involved in these complex mixtures, we employed a variety of traditional methods but also new methods such as the fluctuation theory of Kirkwood and Buff and ab initio quantum mechanical techniques. The Kirkwood-Buff (KB) theory is a rigorous formalism which is free of any of the approximations usually used in the thermodynamic treatment of multicomponent systems. This theory appears to be very fruitful when applied to the above-mentioned difficult systems.

Structure and Dynamics of Solutions Hideyuki Ohtaki, H. Yamatera, 2013-10-22 Recent advances in the study of structural and dynamic properties of solutions have provided a molecular picture of solute-solvent interactions. Although the study of thermodynamic as well as electronic properties of solutions has played a role in the development of research on the rate and mechanism of chemical reactions, such macroscopic and microscopic properties are insufficient for a deeper understanding of fast chemical and biological reactions. In order to fill the gap between the two extremes, it is necessary to know how molecules are arranged in solution and how they change their positions in both the short and long range. This book has been designed to meet these criteria. It is possible to develop a sound microscopic picture for reaction dynamics in solution without molecular-level knowledge of how reacting ionic or neutral species are solvated and how rapidly the molecular environment is changing with time. A variety of actual examples is given as to how and when modern molecular approaches can be used to solve specific solution problems. The following tools are discussed: x-ray and neutron diffraction, EXAFS and XANES, molecular dynamics, and Monte Carlo computer simulations. Raman, infrared, NMR fluorescence, and photoelectron emission spectroscopic methods, conductance, and viscosity measurements, high-pressure techniques, and statistical mechanics methods. Static and dynamic properties of ionic solvation, molecular solvation, ion-pair formation, ligand-exchange reactions, and typical organic solvents are useful for bridging the gap between classical thermodynamic studies and modern single-molecule studies in the gas phase. The book will be of interest to solution physical, inorganic, analytical, and structural chemists, as well

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Encyclopedia of Energy Technology and the Environm, 4 Volume Set Attilio Bisio, Sharon Boots, 1995-04-03 This award winning four volume set examines the impact of energy production technologies on the environment In 235 articles the A to Z work covers such topics as acid rain air pollution aircraft fuel building systems coal combustion computer applications for energy efficient systems risk assessment solar heating waste management planning water power and more This first in the Wiley Encyclopedia Series in Environmental Science this valuable resource features extensive illustration photographs tables and a list of environmental and conversion organizations

Modeling with Differential Equations in Chemical Engineering Stanley M. Walas, 1991 Modelling with Differential Equations in Chemical Engineering covers the modelling of rate processes of engineering in terms of differential equations While it includes the purely mathematical aspects of the solution of differential equations the main emphasis is on the derivation and solution of major equations of engineering and applied science Methods of solving differential equations by analytical and numerical means are presented in detail with many solved examples and problems for solution by the reader Emphasis is placed on numerical and computer methods of solution A key chapter in the book is devoted to the principles of mathematical modelling These principles are applied to the equations in important engineering areas The major disciplines covered are thermodynamics diffusion and mass transfer heat transfer fluid dynamics chemical reactions and automatic control These topics are of particular value to chemical engineers but also are of interest to mechanical civil and environmental engineers as well as applied scientists The material is also suitable for undergraduate and beginning graduate students as well as for

review by practising engineers **Author - Title Catalog** University of California, Berkeley. Library,1963 Elements of Environmental Engineering Kalliat T. Valsaraj,1995-06-27 This text reference covers chemical thermodynamics and reaction kinetics It brings together the components of air water soil sediment and biota and the exchange and transformations that occur in and between them It provides students and researchers with the basic information they need to understand issues in environmental engineering In addition it lays the foundation for more advanced topics in fate and transport modeling and waste treatment and elimination This text also serves as an excellent reference for environmental chemists and toxicologists who need to estimate parameters of various chemicals *Molecular Thermodynamics of Some Highly Asymmetric Liquid Mixtures* Jianzhong Wu,1998

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