

## Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O<sub>2</sub> 1.

How many moles of oxygen are produced from twelve moles of Potassium chlorate?

21 mol O<sub>2</sub> 2.

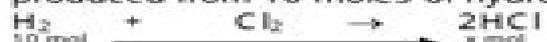
How many moles of oxygen are produced from 14 moles of Potassium chlorate?

45 mol H<sub>2</sub>O 3.

How many moles of water are produced from 22.5 moles of oxygen?

20 mol HCl 4.

Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

21 mol Cl<sub>2</sub> 5.

Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



# Honors Chemistry Unit 6 Test Stoichiometry Practice Test

**Staley Wilbraham (Matta, Waterman)**

**Honors Chemistry Unit 6 Test Stoichiometry Practice Test:**

*Addison-Wesley Chemistry*, 1990      **Catalogs of Courses** University of California, Berkeley, 1990 Includes general and summer catalogs issued between 1878 1879 and 1995 1997      General Catalog Iowa State University, 1989      Pearson Chemistry Unit 6 Staley Wilbraham (Matta, Waterman), 2022

## Whispering the Secrets of Language: An Emotional Journey through **Honors Chemistry Unit 6 Test Stoichiometry Practice Test**

In a digitally-driven earth where screens reign great and immediate interaction drowns out the subtleties of language, the profound secrets and mental nuances concealed within words often get unheard. However, situated within the pages of **Honors Chemistry Unit 6 Test Stoichiometry Practice Test** a interesting fictional prize pulsating with fresh emotions, lies an extraordinary journey waiting to be undertaken. Written by a skilled wordsmith, this charming opus invites readers on an introspective trip, softly unraveling the veiled truths and profound influence resonating within the fabric of each and every word. Within the mental depths of the poignant review, we will embark upon a sincere exploration of the book is core styles, dissect their captivating publishing model, and succumb to the strong resonance it evokes heavy within the recesses of readers hearts.

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