

Fuel Cells Harvest Chemical Energy

6

The real world applications for fuel cells are still in development and require for many reasons cell design that, beyond standard challenges associated with a complex process, address the requirements for development. Fuel cells are made up of an anode, a cathode, and an electrolyte. The anode is the negative terminal and the cathode is the positive terminal. The electrolyte is the medium through which the ions flow. The anode and cathode are separated by a membrane that allows only the ions to pass through, and is often permeable to water. The typical operation of a fuel cell is shown in the diagram below.

Applying Your Knowledge

Answer the questions below and complete additional responses to the questions below.

1. What are the two main components of a fuel cell? How do they work together to produce electricity? (2 points)

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ML Morrison



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