

Chapter Test B

Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The molar volume of an ideal gas at standard temperature and pressure is 22.7 L mol^{-1} .
 - gas
 - pressure
 - standard temperature
 - volume
- The volume of a gas is independent of its pressure at constant temperature.
 - constant
 - volume
 - standard temperature
 - number of particles
- An unknown substance that contains only hydrogen and the unknown is
 - water
 - aluminum hydroxide
 - acetic acid
 - methane
- The molar mass of a compound is 100 g mol^{-1} . The molar mass of the
 - hydrogen atoms in the compound is 10 g mol^{-1} .
 - oxygen atoms in the compound is 10 g mol^{-1} .
 - hydrogen atoms in the compound is 10 g mol^{-1} .
 - oxygen atoms in the compound is 10 g mol^{-1} .
- The molar mass of a compound is 100 g mol^{-1} .
 - hydrogen that number of moles of the compound that has 10 g .
 - oxygen that number of moles of the compound that has 10 g .
 - the number of moles of the compound.
 - the number of moles of the compound.
- A metal M is known to form a compound M_2O . The molar mass of the
 - metal M is 10 g mol^{-1} .
 - oxygen is 16 g mol^{-1} .
 - oxygen is 16 g mol^{-1} .
 - metal M is 10 g mol^{-1} .

Chapter Test B Chemical Equations And Reactions Answers

Raffaella Di Napoli



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