

Writing Ionic Formulas

14

Directions: In each table below, correctly identify the element symbols, charges, and the chemical formula for each ionic compound listed. Once a table is completely correct, the table will turn green!

out of 14

Name:	1. Sodium Chloride			
Symbols:	Na		Cl	
Charges:	+1	-	-1	-
Formula:	NaCl			

Name:	2. Calcium Fluoride			
Symbols:	Ca		F	
Charges:	+2	-	-1	-
Formula:	CaF ₂			

Name:	3. Magnesium Phosphide			
Symbols:	Mg		P	
Charges:	+2	-	-3	-
Formula:	Mg ₃ P ₂			

Name:	4. Beryllium Iodide			
Symbols:	Be		I	
Charges:	+2	-	-1	-
Formula:	BeI ₂			

Name:	5. Sodium Oxide			
Symbols:	Na		O	
Charges:	+1	-	-2	-
Formula:	Na ₂ O			

Name:	6. Aluminum Sulfide			
Symbols:	Al		S	
Charges:	+3	-	-2	-
Formula:	Al ₂ S ₃			

Name:	7. Copper (I) Bromide			
Symbols:	Cu		Br	
Charges:	+1	-	-1	-
Formula:	CuBr			

Name:	8. Sodium Nitrate			
Symbols:	Na		NO ₃	
Charges:	+1	-	-1	-
Formula:	NaNO ₃			

Name:	9. Potassium Sulfate			
Symbols:	K		SO ₄	
Charges:	+1	-	-2	-
Formula:	K ₂ SO ₄			

Name:	10. Ammonium Hydroxide			
Symbols:	NH ₄		OH	
Charges:	+1	-	-1	-
Formula:	NH ₄ OH			

Chemfax Ionic Formula Writing Kit Answers

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Table of Contents Chemfax Ionic Formula Writing Kit Answers

1. Understanding the eBook Chemfax Ionic Formula Writing Kit Answers
 - The Rise of Digital Reading Chemfax Ionic Formula Writing Kit Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemfax Ionic Formula Writing Kit Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemfax Ionic Formula Writing Kit Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemfax Ionic Formula Writing Kit Answers
 - Personalized Recommendations

- Chemfax Ionic Formula Writing Kit Answers User Reviews and Ratings
- Chemfax Ionic Formula Writing Kit Answers and Bestseller Lists
- 5. Accessing Chemfax Ionic Formula Writing Kit Answers Free and Paid eBooks
 - Chemfax Ionic Formula Writing Kit Answers Public Domain eBooks
 - Chemfax Ionic Formula Writing Kit Answers eBook Subscription Services
 - Chemfax Ionic Formula Writing Kit Answers Budget-Friendly Options
- 6. Navigating Chemfax Ionic Formula Writing Kit Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemfax Ionic Formula Writing Kit Answers Compatibility with Devices
 - Chemfax Ionic Formula Writing Kit Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemfax Ionic Formula Writing Kit Answers
 - Highlighting and Note-Taking Chemfax Ionic Formula Writing Kit Answers
 - Interactive Elements Chemfax Ionic Formula Writing Kit Answers
- 8. Staying Engaged with Chemfax Ionic Formula Writing Kit Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemfax Ionic Formula Writing Kit Answers
- 9. Balancing eBooks and Physical Books Chemfax Ionic Formula Writing Kit Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemfax Ionic Formula Writing Kit Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemfax Ionic Formula Writing Kit Answers
 - Setting Reading Goals Chemfax Ionic Formula Writing Kit Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemfax Ionic Formula Writing Kit Answers
 - Fact-Checking eBook Content of Chemfax Ionic Formula Writing Kit Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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