

NAME: _____

Unit 1 Practice Exam 1

1. Atom Structure

Which of the following statements about atomic structure is **NOT** true?

- a) Most of the volume of an atom is empty space.
- b) Electrons are the fundamental particle with a negative charge.
- c) The number of protons and neutrons in an atom must be equal. **correct**
- d) Protons are about 2000 times larger than electrons.

2. Question Functional Relationships

Which of the following statements about functional relationship is **NOT** true?

- a) Plotting (1,2), (2,4), (3,6) is an inverse function. **correct**
- b) Plotting (1,1), (2,4), (3,9) is a parabolic function.
- c) Plotting (1,2), (2,4), (3,8), (4,16) is an exponential function.
- d) Plotting (1,1), (2,0.25), (3,0.11) is an inverse square function.

3. Scientific Notation Math

What is $[(8 \times 10^{-5})(3 \times 10^9)]/(4 \times 10^{-5})$?

- a) 6×10^9 **correct**
- b) 6×10^{-1}
- c) 9.6×10^{10}
- d) 6.6×10^9

4. Approximations

The atomic weights of the halogens are listed below. If you rounded to the nearest integer to perform a calculation of molar mass, which would have the largest error?

- a) F: 19.0 g/mol
- b) Cl: 35.45 g/mol **correct**
- c) Br: 79.9 g/mol
- d) I: 126.9 g/mol

Chemistry Midterm Lab Answers

VM Jensen



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Table of Contents Chemistry Midterm Lab Answers

1. Understanding the eBook Chemistry Midterm Lab Answers
 - The Rise of Digital Reading Chemistry Midterm Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Midterm Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Midterm Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Midterm Lab Answers
 - Personalized Recommendations
 - Chemistry Midterm Lab Answers User Reviews and Ratings

- Chemistry Midterm Lab Answers and Bestseller Lists
- 5. Accessing Chemistry Midterm Lab Answers Free and Paid eBooks
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 - Highlighting and Note-Taking Chemistry Midterm Lab Answers
 - Interactive Elements Chemistry Midterm Lab Answers
- 8. Staying Engaged with Chemistry Midterm Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Midterm Lab Answers
- 9. Balancing eBooks and Physical Books Chemistry Midterm Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Midterm Lab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Midterm Lab Answers
 - Setting Reading Goals Chemistry Midterm Lab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Midterm Lab Answers
 - Fact-Checking eBook Content of Chemistry Midterm Lab Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
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