

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 3

Q1. In a hydrocarbon, the mass ratio of hydrogen and carbon is 1:3; the empirical formula of hydrocarbon is _____.

- a.) CH_4
- b.) CH_2
- c.) C_2H
- d.) CH_3

Q2. 80g of oxygen contains as many atoms as in _____.

- a.) 10 g of hydrogen
- b.) 5 g of hydrogen
- c.) 8 g of hydrogen
- d.) 1 g of hydrogen

Q3. A mixture of gases contains H_2 and O_2 gases in the ratio 1:4. What is the molar ratio of the two gases in the mixture?

- a.) 16:1
- b.) 2:1
- c.) 1:4
- d.) 4:1

Q4. Which of the following concentration terms is/are independent of temperature?

- a.) Molality
- b.) Mole fraction
- c.) Molarity
- d.) Normality

Q5. Which of the following contains the highest number of atoms?

- a.) 1.0 g of water
- b.) 1.0 g of silver
- c.) 1.0 g of nitrogen
- d.) 1.0 g of propane

Chemistry Chapter 1 Worksheet

AW Rasmussen



Chemistry Chapter 1 Worksheet:

Decoding **Chemistry Chapter 1 Worksheet**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Chapter 1 Worksheet**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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