

0620 Chemistry Past Papers (Year 2022) – Paper 2 MCQs

Feb/Mar 2022 – 22

1. Which gas has the fastest rate of diffusion?
- A H_2 B CH_4 C CO_2 D SO_2
2. In which state does 1 dm^3 of methane contain the most particles?
- A gas at 100°C
B gas at room temperature
C liquid
D solid
3. Which statement explains why isotopes of the same element have the same chemical properties?
- A They have the same electronic structure.
B They have the same relative mass.
C They have the same nucleon number.
D They have the same proton number.

4. The electronic structures of atoms P and Q are shown.



P and Q form an ionic compound.

What is the formula of the compound?

- A PQ B P_2Q C P_2Q_3 D PQ_3
5. Fermentation of sugar produces a mixture of ethanol solution and solid yeast.
- How is the solid yeast removed from the mixture?
- A crystallisation
B distillation
C filtration
D fractional distillation
6. Compound X contains carbon, hydrogen and oxygen only.
- By mass, it contains 26.7% carbon and 2.2% hydrogen.
- What is the empirical formula of X?

- A CHO B C_2HO C CH_2O D CHO_2

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1. Understanding the eBook Chemistry 0620 Jun 2001
 - The Rise of Digital Reading Chemistry 0620 Jun 2001
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 0620 Jun 2001
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 0620 Jun 2001
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 0620 Jun 2001

- Personalized Recommendations
 - Chemistry 0620 Jun 2001 User Reviews and Ratings
 - Chemistry 0620 Jun 2001 and Bestseller Lists
5. Accessing Chemistry 0620 Jun 2001 Free and Paid eBooks
 - Chemistry 0620 Jun 2001 Public Domain eBooks
 - Chemistry 0620 Jun 2001 eBook Subscription Services
 - Chemistry 0620 Jun 2001 Budget-Friendly Options
 6. Navigating Chemistry 0620 Jun 2001 eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 0620 Jun 2001 Compatibility with Devices
 - Chemistry 0620 Jun 2001 Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 0620 Jun 2001
 - Highlighting and Note-Taking Chemistry 0620 Jun 2001
 - Interactive Elements Chemistry 0620 Jun 2001
 8. Staying Engaged with Chemistry 0620 Jun 2001
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 0620 Jun 2001
 9. Balancing eBooks and Physical Books Chemistry 0620 Jun 2001
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 0620 Jun 2001
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry 0620 Jun 2001
 - Setting Reading Goals Chemistry 0620 Jun 2001
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry 0620 Jun 2001

- Fact-Checking eBook Content of Chemistry 0620 Jun 2001
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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